

# **Towards ideal topological materials: Comprehensive database searches using symmetry indicators**

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# Cooperation

**Feng Tang;** *Nanjing University*

**Di Wang;** *Nanjing University*



**Hoi Chun Po & Ashvin Vishwanath;** *Harvard University*



*Tang, Po, Vishwanath, Wan\**, *Nature Physics* **15**, 470 (2019)

*Tang, Po, Vishwanath, Wan\**, *Science Advances* **5**, eaau8725 (2019)

*Tang, Po, Vishwanath, Wan\**, *Nature* **566**, 486 (2019)

*Wang, Tang, Ji, Zhang, Vishwanath, Po, Wan\**, *arXiv:1906.01283* (2019)

# Content

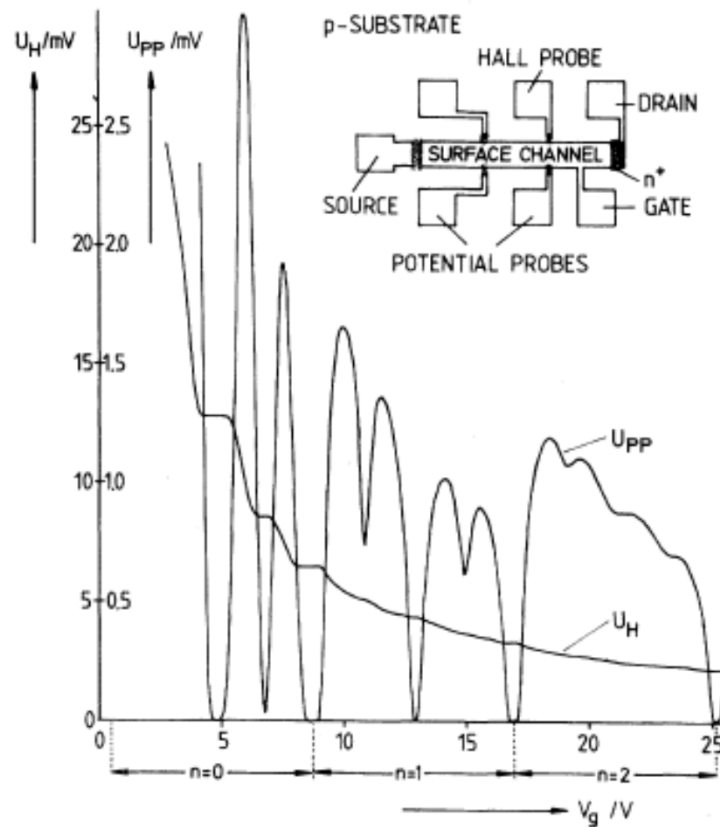
Symmetry Indicators

Topological Materials

# 拓扑

➤ 朗道相变理论 → 对称性、序参量

➤ 1980





## Quantized Hall Conductance in a Two-Dimensional Periodic Potential

D. J. Thouless, M. Kohmoto,<sup>(a)</sup> M. P. Nightingale, and M. den Nijs

*Department of Physics, University of Washington, Seattle, Washington 98195*

(Received 30 April 1982)

The Hall conductance of a two-dimensional electron gas has been studied in a uniform magnetic field and a periodic substrate potential  $U$ . The Kubo formula is written in a form that makes apparent the quantization when the Fermi energy lies in a gap. Explicit expressions have been obtained for the Hall conductance for both large and small  $U/\hbar\omega_c$ .

$$\sigma_H = \frac{ie^2}{A_0 \hbar} \sum_{\epsilon_\alpha < E_F} \sum_{\epsilon_\beta > E_F} \frac{(\partial \hat{H} / \partial k_1)_{\alpha\beta} (\partial \hat{H} / \partial k_2)_{\beta\alpha} - (\partial \hat{H} / \partial k_2)_{\alpha\beta} (\partial \hat{H} / \partial k_1)_{\beta\alpha}}{(\epsilon_\alpha - \epsilon_\beta)^2},$$

# Chern Number

- 2D BZ是一个torus, 它组成的曲面包含的monopole的个数只能是整数

$$\sigma_H = \frac{ie^2}{A_0 \hbar} \sum_{\epsilon_\alpha < E_F} \sum_{\epsilon_\beta > E_F} \frac{(\partial \hat{H} / \partial k_1)_{\alpha\beta} (\partial \hat{H} / \partial k_2)_{\beta\alpha} - (\partial \hat{H} / \partial k_2)_{\alpha\beta} (\partial \hat{H} / \partial k_1)_{\beta\alpha}}{(\epsilon_\alpha - \epsilon_\beta)^2}$$

$$\Omega_{\mu\nu}^n(\mathbf{R}) = i \sum_{n' \neq n} \frac{\langle n | \partial H / \partial R^\mu | n' \rangle \langle n' | \partial H / \partial R^\nu | n \rangle - (\nu \leftrightarrow \mu)}{(\epsilon_n - \epsilon_{n'})^2}.$$

$$\sigma_{xy} = \frac{e^2}{\hbar} \int_{\text{BZ}} \frac{d^2 k}{(2\pi)^2} \Omega_{k_x k_y} = n \frac{e^2}{h}.$$

Chern Number

# Z2 Topological insulator

PRL **95**, 226801 (2005)

PHYSICAL REVIEW LETTERS

week ending  
25 NOVEMBER 2005

## Quantum Spin Hall Effect in Graphene

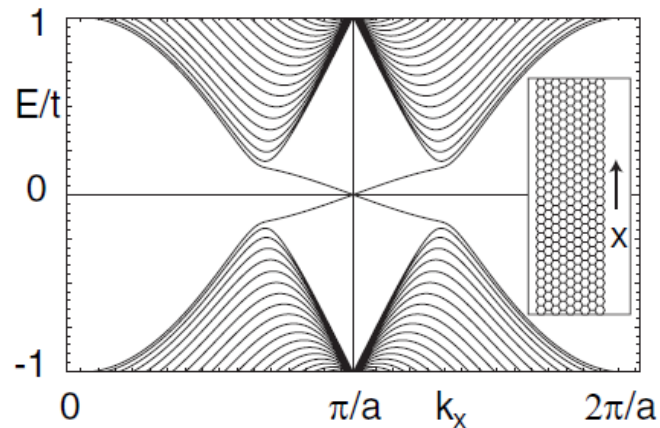
C. L. Kane and E. J. Mele

PRL **95**, 146802 (2005)

PHYSICAL REVIEW LETTERS

30

## $Z_2$ Topological Order and the Quantum Spin Hall Effect



C. L. Kane and E. J. Mele

$Z_2$ 拓扑不变量

时间反演对称

**SOC**

# Topological invariants in energy band

- **Chern number** in  $T$ -symmetry breaking system
- **$Z_2$  topological invariant** in  $T$ -symmetry invariant system

# Relativistic Quantum Mechanics

➤ Klein-Gordon equation 1926

相对论+量子化

$$\begin{array}{l}
 \mathbf{k} \rightarrow -i\nabla \\
 E \rightarrow i\partial_t
 \end{array}
 + E^2 = m^2 + p^2
 \quad \Rightarrow \quad
 \begin{array}{l}
 (\square - \mu^2)\phi(\mathbf{x}, t) = 0 \\
 \square = \nabla^2 - \partial_t^2
 \end{array}$$

*Negative Energies*  $e^{(i\mathbf{k}\cdot\mathbf{x} - iEt)}$   $\Rightarrow E = \pm(\mathbf{k}^2 + \mu^2)^{1/2}$

$$\begin{array}{l}
 \phi^*(\square - \mu^2)\phi = 0 \\
 \phi(\square - \mu^2)\phi^* = 0
 \end{array}
 \Rightarrow
 \begin{array}{l}
 \partial_\mu j_\mu(x) = 0 \quad \rho(x) = 2E\phi^*(x)\phi(x) \\
 \text{non-positive-definite} \\
 j_\mu(x) = i[(\partial_\mu\phi^*(x))\phi(x) - \phi^*(x)\partial_\mu\phi(x)]
 \end{array}$$

# Relativistic Quantum Mechanics

➤ Dirac equation 1928

$$(\gamma_\mu \partial_\mu + m)\psi = 0$$

$$\gamma_\mu \gamma_\nu + \gamma_\nu \gamma_\mu = 2\delta_{\mu\nu}$$

$$(\mu, \nu = 1 \dots 4)$$

➤ Dirac representation

Euclidian vector  $x_\mu = (\mathbf{x}, it)$

Greek letters  $\mu, \nu, \dots$  take values from 1 to  $d+1$

$$A \cdot B = \sum_{\mu} A_{\mu} B_{\mu}$$

$$c = \hbar = 1$$

# Relativistic Quantum Mechanics

➤ Dirac equation 1928

$$\gamma_\mu \gamma_\nu + \gamma_\nu \gamma_\mu = 2\delta_{\mu\nu}$$

$$(\gamma_\mu \partial_\mu + m)\psi = 0$$

$$(\mu, \nu = 1 \dots 4)$$

➤ Dirac representation

在静止系，Dirac方程对角化

spinor index as  $(\xi\sigma)$  where  $\xi = \pm$  and  $\sigma = \uparrow, \downarrow$

➤ Weyl representation

the representation of the Lorentz group is  $(\frac{1}{2}, 0) \oplus (0, \frac{1}{2})$

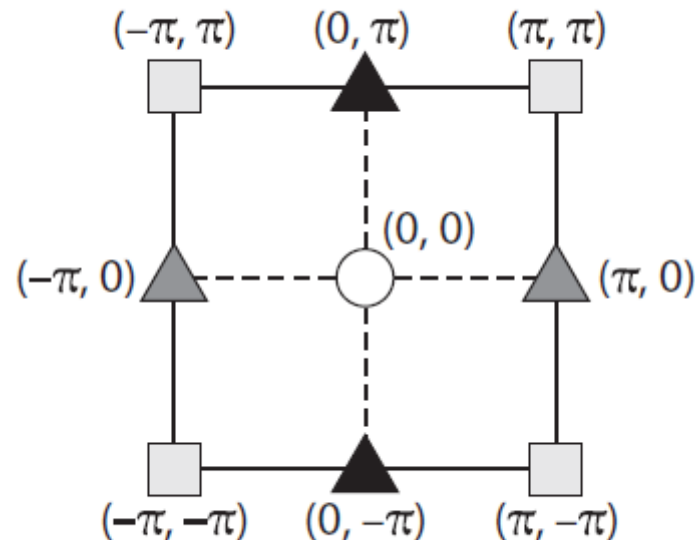
spinor index  $(\zeta\sigma)$  where  $\zeta = L, R$  and  $\sigma = \uparrow, \downarrow$

$$\psi_{+,\sigma} = \frac{1}{\sqrt{2}}(\psi_{L,\sigma} - \psi_{R,\sigma}), \quad \psi_{-,\sigma} = \frac{1}{\sqrt{2}}(\psi_{L,\sigma} + \psi_{R,\sigma})$$

# Topological Insulator

- Bulk insulator
- Topological protected surface state
- Mass term in Dirac Equation change Sign

$$H_t = \alpha \cdot \mathbf{p} + \beta m$$





# Bulk-boundary correspondence

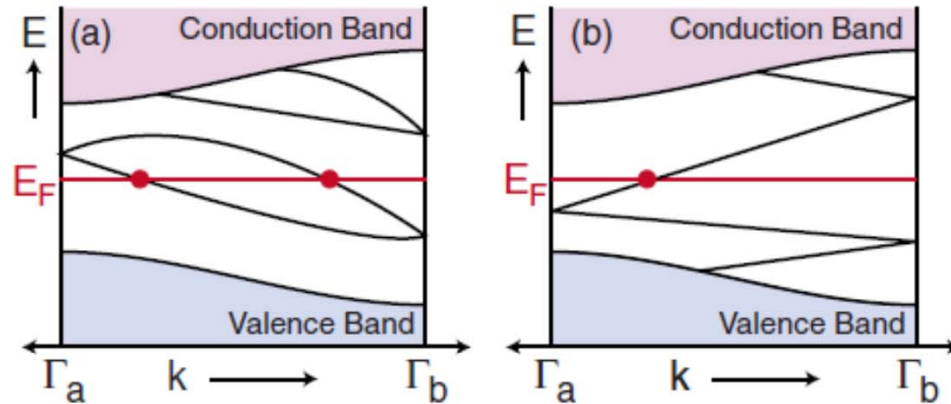


FIG. 3. (Color online) Electronic dispersion between two boundary Kramers degenerate points  $\Gamma_a=0$  and  $\Gamma_b=\pi/a$ . In (a) the number of surface states crossing the Fermi energy  $E_F$  is even, whereas in (b) it is odd. An odd number of crossings leads to topologically protected metallic boundary states.

**Weak topological insulator**

$Z_2$  invariants  $(\gamma_0; \gamma_1, \gamma_2, \gamma_3)$

**Strong topological insulator**

$$\mathcal{H}_{\text{surface}} = -i\hbar v_F \vec{\sigma} \cdot \vec{\nabla},$$

# Justify 3D TI

- ① Surface State Electron Structure
- ② Adiabatic Continue Band Transformation
- ③ Band inversion
- ④ Calculate  $Z_2$

# Calculate $Z_2$

➤ **With inversion symmetry**

Fu-Kane theorem (*Fu & Kane, 2007*)

➤ **Without inversion symmetry**

directly calculate  $Z_2$

(*Feng, Wen, Zhou, Xiao & Yao (2012)*)

$$Z_2 = \frac{1}{2\pi} \left[ \oint_{\partial\mathcal{B}^+} dk \cdot \mathcal{A}(k) - \int_{\mathcal{B}^+} d^2k \mathcal{F}(k) \right] \text{ mod } 2,$$

where  $\mathcal{A}(k)$  and  $\mathcal{F}(k)$  are the Berry connection and Berry curvature, respectively,

$$\mathcal{A}(k) = i \sum_n \langle u_n(k) | \nabla_k u_n(k) \rangle \quad \mathcal{F}(k) = \nabla_k \times \mathcal{A}(k) \Big|_z.$$

# 周期表

PHYSICAL REVIEW B 78, 195125 (2008)

## Classification of topological insulators and superconductors in three spatial dimensions

Andreas P. Schnyder,<sup>1</sup> Shinsei Ryu,<sup>1</sup> Akira Furusaki,<sup>2</sup> and Andreas W. W. Ludwig<sup>3</sup>

		TRS	PHS	SLS	$d=1$	$d=2$	$d=3$
Standard (Wigner-Dyson)	A (unitary)	0	0	0	-	$\mathbb{Z}$	-
	AI (orthogonal)	+1	0	0	-	-	-
	AII (symplectic)	-1	0	0	-	$\mathbb{Z}_2$	$\mathbb{Z}_2$
Chiral (sublattice)	AIII (chiral unitary)	0	0	1	$\mathbb{Z}$	-	$\mathbb{Z}$
	BDI (chiral orthogonal)	+1	+1	1	$\mathbb{Z}$	-	-
	CII (chiral symplectic)	-1	-1	1	$\mathbb{Z}$	-	$\mathbb{Z}_2$
BdG	D	0	+1	0	$\mathbb{Z}_2$	$\mathbb{Z}$	-
	C	0	-1	0	-	$\mathbb{Z}$	-
	DIII	-1	+1	1	$\mathbb{Z}_2$	$\mathbb{Z}_2$	$\mathbb{Z}$
	CI	+1	-1	1	-	-	$\mathbb{Z}$

TABLE I. Ten symmetry classes of single-particle Hamiltonians classified in terms of the presence or absence of time-reversal symmetry (TRS) and particle-hole symmetry (PHS), as well as “sublattice” (or “chiral”) symmetry (SLS) (Refs. 37 and 38). In the table, the absence of symmetries is denoted by “0.” The presence of these symmetries is denoted by either “+1” or “-1,” depending on whether the (antiunitary) operator implementing the symmetry at the level of the single-particle Hamiltonian squares to “+1” or “-1” (see text). [The index  $\pm 1$  equals  $\eta_c$  in Eq. (1b); here  $\epsilon_c = +1$  and  $-1$  for TRS and PHS, respectively.] For the first six entries of the table (which can be realized in nonsuperconducting systems), TRS = +1 when the SU(2) spin is an integer [called TRS (even) in the text] and TRS = -1 when it is a half-integer [called TRS (odd) in the text]. For the last four entries, the superconductor “Bogoliubov–de Gennes” (BdG) symmetry classes D, C, DIII, and CI, the Hamiltonian preserves SU(2) spin-1/2 rotation symmetry when PHS = -1 [called PHS (singlet) in the text], while it does not preserve SU(2) when PHS = +1 [called PHS (triplet) in the text]. The last three columns list all topologically non-trivial quantum ground states as a function of symmetry class and spatial dimension. The symbols  $\mathbb{Z}$  and  $\mathbb{Z}_2$  indicate whether the space of quantum ground states is partitioned into topological sectors labeled by an integer or a  $\mathbb{Z}_2$  quantity, respectively.

# Topological Crystalline Insulators

## Spatial Symmetries

PRL **106**, 106802 (2011)

PHYSICAL REVIEW LETTERS

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### Topological Crystalline Insulators

Liang Fu

ARTICLES

PUBLISHED ONLINE: 16 DECEMBER 2012 | DOI: 10.1038/NPHYS2513

nature  
physics

## The space group classification of topological band-insulators

Robert-Jan Slager<sup>1</sup>, Andrej Mesaros<sup>2</sup>, Vladimir Juričić<sup>1\*</sup> and Jan Zaanen<sup>1</sup>

Y. Ando and L. Fu, *Annual Review of Condensed Matter Physics* **6**, 361 (2015).

# Topological Crystalline Insulators

## ➤ Mirror Chern Insulator

*Ando and Fu, Annu. Rev. Condens. Matter Phys. 6, 361 (2015)*

## ➤ Hourglass Fermions

*Wang, Alexandradinata, Cava, Bernevig, Nature 532, 189 (2016).*

## ➤ Wallpaper Fermions

*Wieder et al., Science 361, 246 (2018)*

## ➤ Higher-order topological insulators

*Fang&Fu; Schindler et al., Langbehn et al., Song, Fang&Fang; Benalcazar et al., (2017)*

...

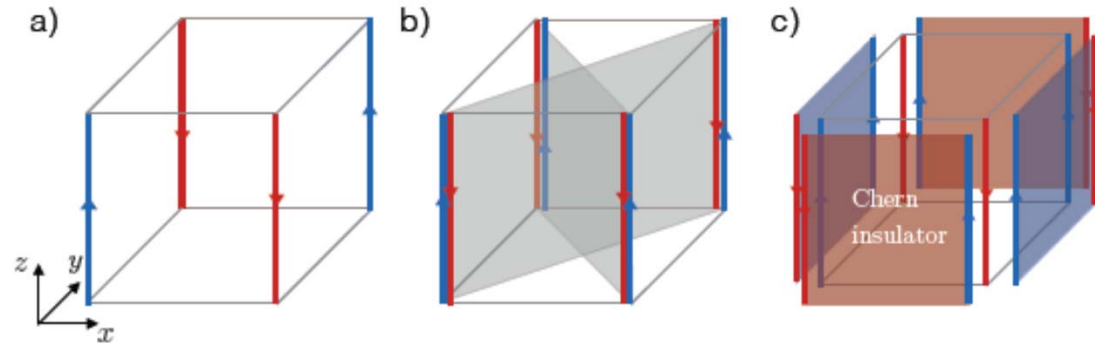
# Higher-Order Topological Insulators

arXiv:1708.03636

Frank Schindler,<sup>1</sup> Ashley M. Cook,<sup>1</sup> Maia G. Vergniory,<sup>2,3</sup> Zhijun Wang,<sup>4</sup>  
Stuart S. P. Parkin,<sup>5</sup> B. Andrei Bernevig,<sup>4,2,6</sup> and Titus Neupert<sup>1</sup>

bulk  $(d) \rightarrow d-1, d-2, d-3$

Second order



## Topological Insulators Turn a Corner

Theorists have discovered topological insulators that are insulating in their interior and on their surfaces but have conducting channels at corners or along edges.

by Siddharth A. Parameswaran\* and Yuan Wan

- [2] W. A. Benalcazar, B. A. Bernevig, and T. L. Hughes, "Quantized Electric Multipole Insulators," *Science* **357**, 61 (2017).
- [3] W. A. Benalcazar, B. A. Bernevig, and T. L. Hughes, "Electric Multipole Moments, Topological Multipole Moment Pumping, and Chiral Hinge States in Crystalline Insulators," *Phys. Rev. B* **96**, 245115 (2017).
- [4] Z. Song, Z. Fang, and C. Fang, " $(d - 2)$ -dimensional edge states of rotation symmetry protected topological states," *Phys. Rev. Lett.* **119**, 246402 (2017).
- [5] J. Langbehn, Y. Peng, L. Trifunovic, F. von Oppen, and P. W. Brouwer, "Reflection-symmetric second-order topological insulators and superconductors," *Phys. Rev. Lett.* **119**, 246401 (2017).

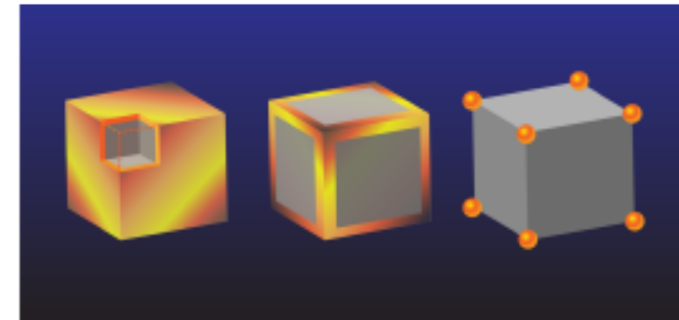


Figure 1: Usually, 3D topological insulators conduct via gapless states on their 2D surfaces but are insulating in their bulk (left). Recently proposed second- and third-order 3D TIs have gapless states on their 1D hinges (middle) or 0D corners (right), respectively, and they constitute a new class of topological phases of matter. (APS/Alan Stonebraker)

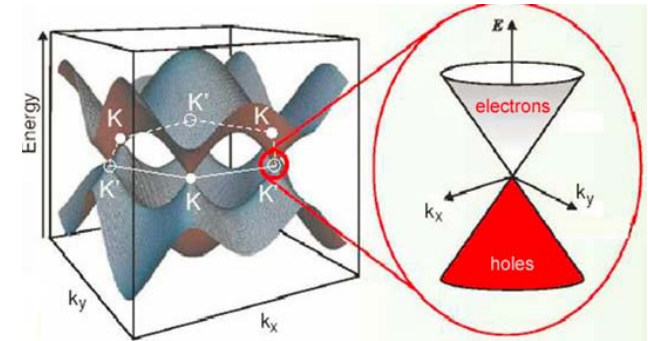


# Dirac Fermion in Condensed Matter

## Massless Dirac Fermion

K-point at BZ of Graphene (1947)

2004, 2005



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## Two-dimensional gas of massless Dirac fermions in graphene

Vol 438|10 November 2005|doi:10.1038/nature04233

K. S. Novoselov<sup>1</sup>, A. K. Geim<sup>1</sup>, S. V. Morozov<sup>2</sup>, D. Jiang<sup>1</sup>, M. I. Katsnelson<sup>3</sup>, I. V. Grigorieva<sup>1</sup>, S. V. Dubonos<sup>2</sup> & A. A. Firsov<sup>2</sup>

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Vol 438|10 November 2005|doi:10.1038/nature04235

## Experimental observation of the quantum Hall effect and Berry's phase in graphene

Yuanbo Zhang<sup>1</sup>, Yan-Wen Tan<sup>1</sup>, Horst L. Stormer<sup>1,2</sup> & Philip Kim<sup>1</sup>



# Dirac Fermion in CM

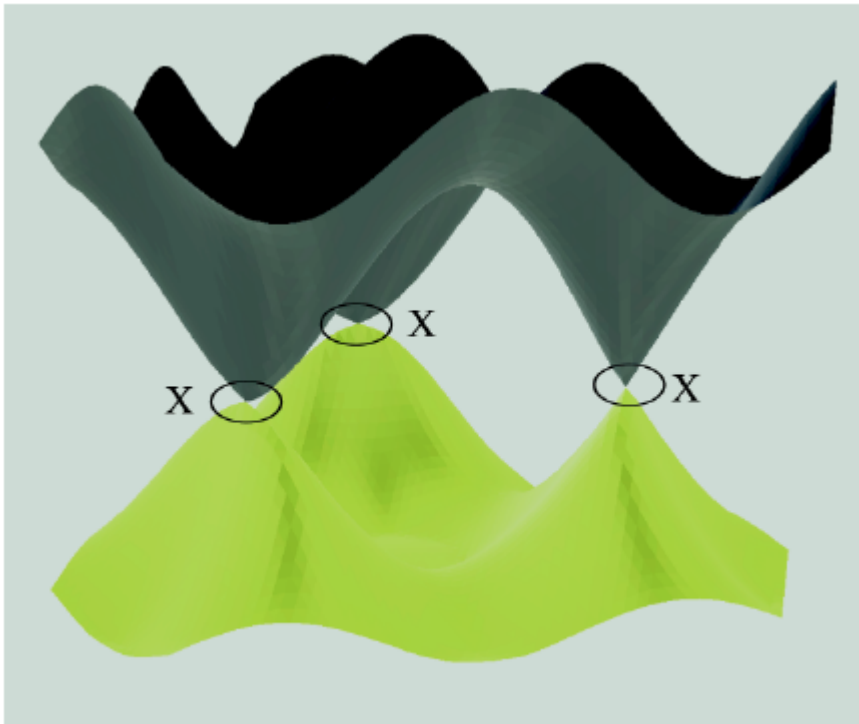
## Massless Dirac Fermion 3D

### symmetry protected degeneracies

Dirac Semimetal in Three Dimensions

PRL **108**, 140405 (2012)

S. M. Young,<sup>1</sup> S. Zaheer,<sup>2</sup> J. C. Y. Teo,<sup>2,\*</sup> C. L. Kane,<sup>2</sup> E. J. Mele,<sup>2</sup> and A. M. Rappe<sup>1</sup>



- **Four dimensional irreducible representations at high symmetry points of BZ**
- **Linearly in all directions around these points**

# Dirac Fermion in CM

Dirac semimetal and topological phase transitions in  $A_3\text{Bi}$  ( $A = \text{Na, K, Rb}$ )

Zhijun Wang,<sup>1</sup> Yan Sun,<sup>2</sup> Xing-Qiu Chen,<sup>2</sup> Cesare Franchini,<sup>2</sup> Gang Xu,<sup>1</sup> Hongming Weng,<sup>1,\*</sup> Xi Dai,<sup>1</sup> and Zhong Fang<sup>1,†</sup>

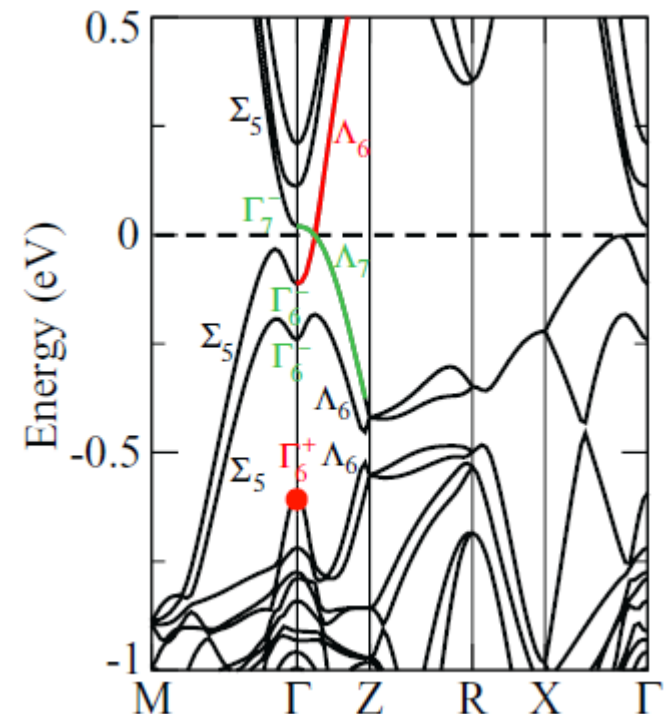
PHYSICAL REVIEW B **85**, 195320 (2012)

Three-dimensional Dirac semimetal and quantum transport in  $\text{Cd}_3\text{As}_2$

Zhijun Wang, Hongming Weng,<sup>\*</sup> Quansheng Wu, Xi Dai, and Zhong Fang<sup>†</sup>

PHYSICAL REVIEW B **88**, 125427 (2013)

- Two bands along high symmetry lines of BZ belong to different two dimensional irreducible representations →

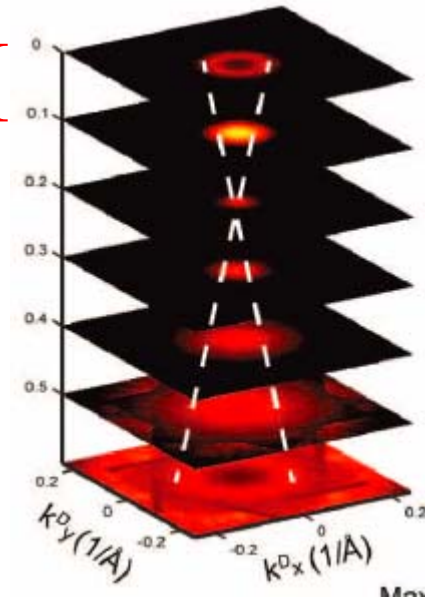


# Dirac Fermion in CM

## Discovery of a Three-Dimensional Topological Dirac Semimetal, Na<sub>3</sub>Bi

Z. K. Liu,<sup>1\*</sup> B. Zhou,<sup>2,3\*</sup> Y. Zhang,<sup>3</sup> Z. J. Wang,<sup>4</sup> H. M. Weng,<sup>4,5</sup> D. Prabhakaran,<sup>2</sup> S.-K. Mo,<sup>3</sup>  
Z. X. Shen,<sup>1</sup> Z. Fang,<sup>4,5</sup> X. Dai,<sup>4,5</sup> Z. Hussain,<sup>3</sup> Y. L. Chen<sup>2,6†</sup>

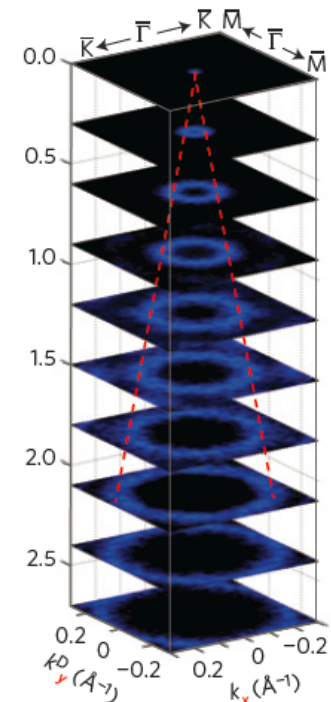
21 FEBRUARY 2014 VOL 343 SCIENCE



## A stable three-dimensional topological Dirac semimetal Cd<sub>3</sub>As<sub>2</sub>

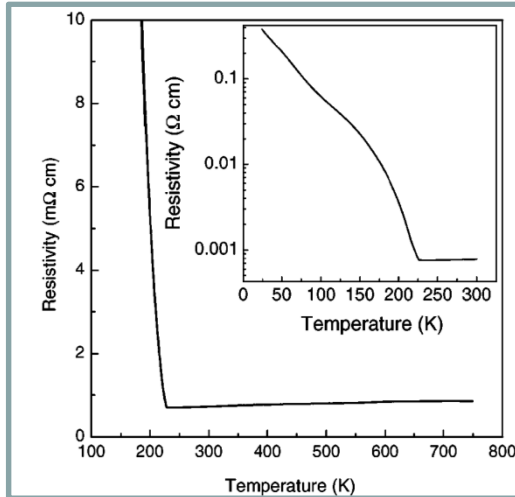
Z. K. Liu<sup>1†</sup>, J. Jiang<sup>2,3†</sup>, B. Zhou<sup>2,4†</sup>, Z. J. Wang<sup>5†</sup>, Y. Zhang<sup>1,4</sup>, H. M. Weng<sup>5</sup>, D. Prabhakaran<sup>2</sup>, S.-K. Mo<sup>4</sup>,  
H. Peng<sup>2</sup>, P. Dudin<sup>6</sup>, T. Kim<sup>6</sup>, M. Hoesch<sup>6</sup>, Z. Fang<sup>5</sup>, X. Dai<sup>5</sup>, Z. X. Shen<sup>1</sup>, D. L. Feng<sup>3</sup>, Z. Hussain<sup>4</sup>  
and Y. L. Chen<sup>1,2,4,6\*</sup>

PUBLISHED ONLINE: 25 MAY 2014 | DOI: 10.1038/NMAT3990



# 磁性体系工作一大难题

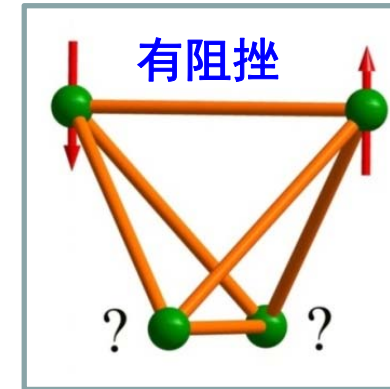
烧绿石晶格结构 $A_2B_2O_7$ 有磁阻挫，定其磁结构是一大难题



有磁结构导致的金属绝缘体转变

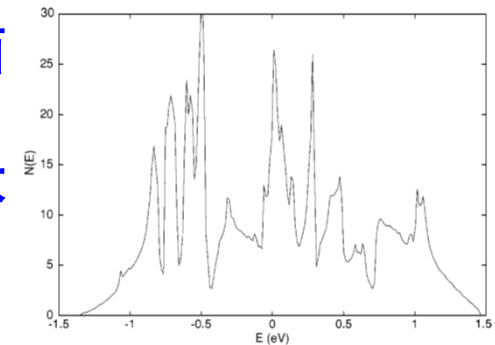
$Cd_2Os_2O_7$

Mandrus *et al.*, PRB 63, 195104 (2001)



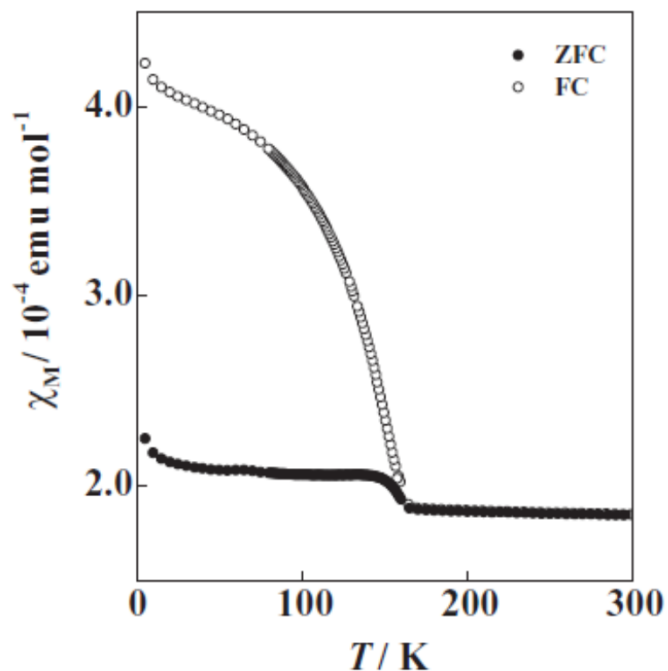
The Os sublattice consists of corner sharing tetrahedra, with the Os atoms at the corners. As a result it is not possible to find a structure in which all nearest-neighbor Os bonds are antiferromagnetic, and additionally, the lattice is strongly geometrically frustrated for nearest-neighbor spin Hamiltonians. However, based on our band-structure results, the proximity of  $Cd_2Os_2O_7$  to magnetism is itinerant in nature,

Os原子在子晶格通过四面体共顶点连接。当最近邻Os反铁磁排列时，有强大的几何阻挫因此不可能发现磁结构



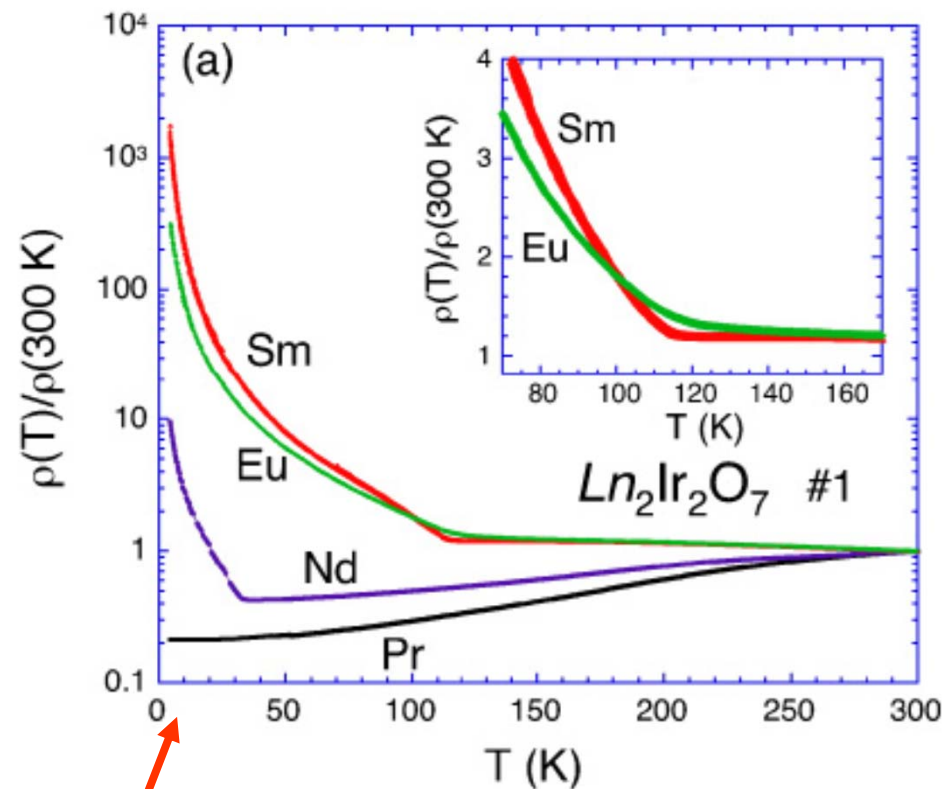
Singh, Blaha, Schwarz, Sofo, PRB 65, 155109 (2002)

# 烧绿石结构铱氧化物 $A_2Ir_2O_7$ ( $A=Y$ , 稀土元素) 实验事实



N. Taira, et al., (2001)

这些材料表现强的磁响应  
实验定磁结构不容易  
基本没有能带计算的工作



MATSUHIRA et al., (2007)

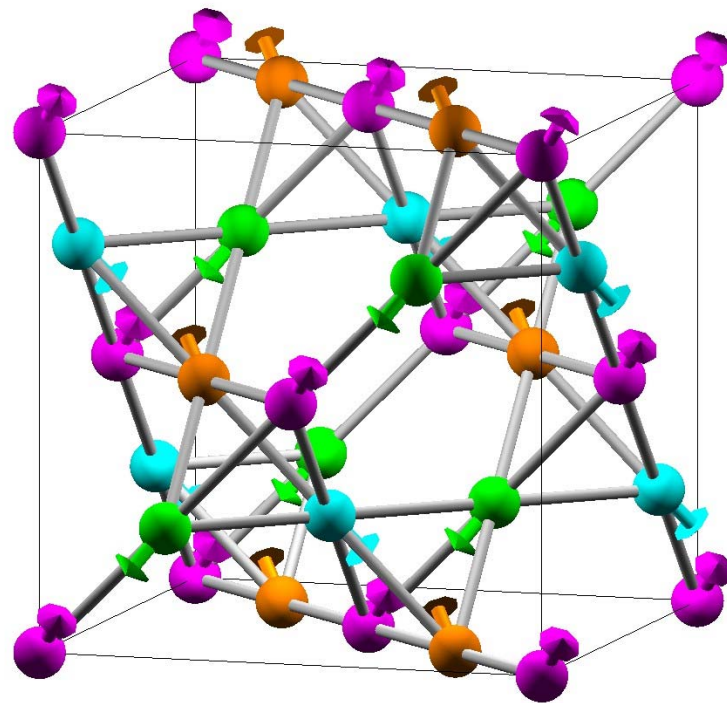
**U !!!**

# 磁性基态构型

## All-in/all-out非共线磁结构

- 根据:

- 1) 有转过去的趋势
- 2) 此构型是我们计算的唯一稳定的磁构型
- 3)  $J(q)$ 在 $q=0$ 是极大
- 4) 没有Fermi surface nesting





# 实验证实

**Magnetic transition, long-range order, and moment fluctuations in the pyrochlore iridate  $\text{Eu}_2\text{Ir}_2\text{O}_7$**

Songrui Zhao,<sup>1,\*</sup> J. M. Mackie,<sup>1</sup> D. E. MacLaughlin,<sup>1</sup> O. O. Bernal,<sup>2</sup> J. J. Ishikawa,<sup>3</sup> Y. Ohta,<sup>3</sup> and S. Nakatsuji<sup>3</sup>

**Magnetic order in the pyrochlore iridates  $A_2\text{Ir}_2\text{O}_7$  ( $A = \text{Y}, \text{Yb}$ )**

S. M. Disseler,<sup>1</sup> Chetan Dhital,<sup>1</sup> A. Amato,<sup>2</sup> S. R. Giblin,<sup>3</sup> Clarina de la Cruz,<sup>4</sup> Stephen D. Wilson,<sup>1</sup> and M. J. Graf<sup>1,†</sup>

**Continuous transition between antiferromagnetic insulator and paramagnetic metal in the pyrochlore iridate  $\text{Eu}_2\text{Ir}_2\text{O}_7$**

Jun J. Ishikawa,<sup>\*</sup> Eoin C. T. O'Farrell, and Satoru Nakatsuji<sup>†</sup>

**Magnetic order and the electronic ground state in the pyrochlore iridate  $\text{Nd}_2\text{Ir}_2\text{O}_7$**

S. M. Disseler,<sup>1</sup> Chetan Dhital,<sup>1</sup> T. C. Hogan,<sup>1</sup> A. Amato,<sup>2</sup> S. R. Giblin,<sup>3</sup> Clarina de la Cruz,<sup>4</sup> A. Daoud-Aladine,<sup>3</sup> Stephen D. Wilson,<sup>1</sup> and M. J. Graf<sup>1</sup>

**Emergence of Magnetic Long-range Order in Frustrated Pyrochlore  $\text{Nd}_2\text{Ir}_2\text{O}_7$  with Metal-insulator Transition**

K. Tomiyasu,<sup>1,\*</sup> K. Matsuhira,<sup>2</sup> K. Iwasa,<sup>1</sup> M. Watahiki,<sup>1</sup> S. Takagi,<sup>2</sup> M. Wakeshima,<sup>3</sup> Y. Hinatsu,<sup>3</sup> M. Yokoyama,<sup>4</sup> K. Ohoyama,<sup>5</sup> and K. Yamada<sup>6</sup>

# All-in/all-out非共线磁结构

PRL **108**, 247204 (2012)

PHYSICAL REVIEW LETTERS

week ending  
15 JUNE 2012

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## Noncollinear Magnetism and Spin-Orbit Coupling in 5d Pyrochlore Oxide $\text{Cd}_2\text{Os}_2\text{O}_7$

We investigate the electronic and magnetic properties of the pyrochlore oxide  $\text{Cd}_2\text{Os}_2\text{O}_7$  using the density-functional theory plus on-site repulsion ( $U$ ) method, and depict the ground-state phase diagram with respect to  $U$ . We conclude that the all-in–all-out noncollinear magnetic order is stable in a wide range of  $U$ . We also show that the easy-axis anisotropy arising from the spin-orbit coupling plays a significant role in stabilizing the all-in–all-out magnetic order. A pseudogap was observed near the transition between

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PRL **108**, 247205 (2012)

PHYSICAL REVIEW LETTERS

week ending  
15 JUNE 2012

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## Tetrahedral Magnetic Order and the Metal-Insulator Transition in the Pyrochlore Lattice of $\text{Cd}_2\text{Os}_2\text{O}_7$

accompanied with any spatial symmetry breaking. We propose a noncollinear all-in–all-out spin arrangement on the tetrahedral network made of Os atoms. Based on this we suggest that the transition is not caused by the Slater mechanism as believed earlier but by an alternative mechanism related to the

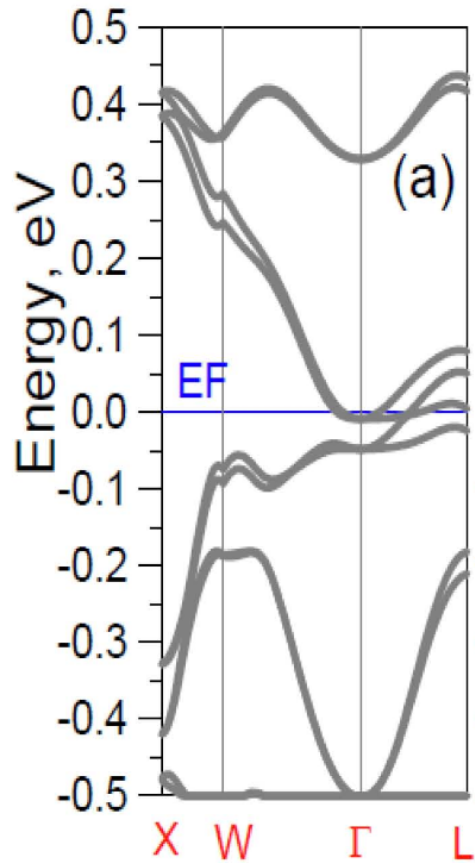


# Correlation $\rightarrow$ band structure evolution

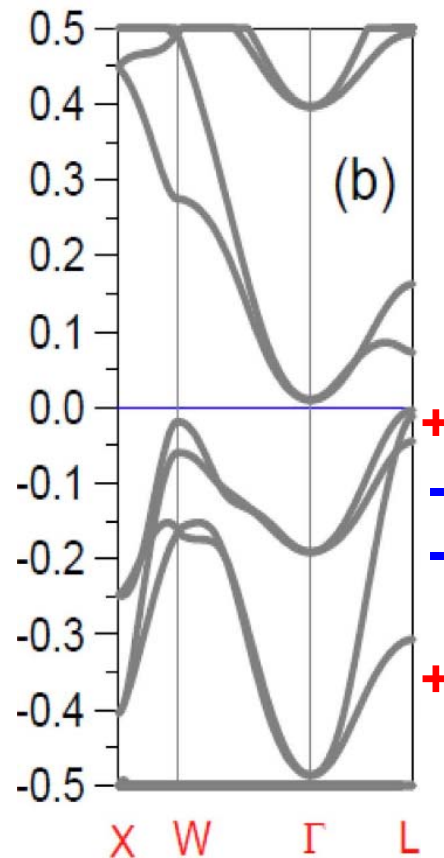
$Y_2Ir_2O_7$

**LDA+U method**

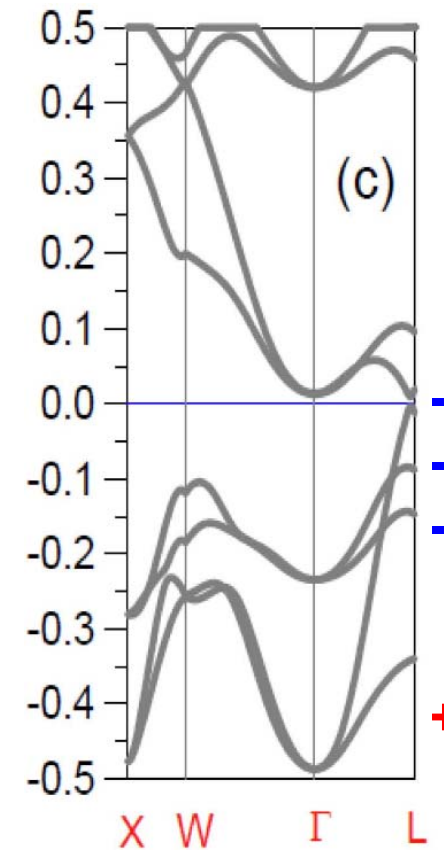
All-in/All-out magnetic order



$U=0$



$U=1.5$  eV



$U=2.0$  eV

enlarge

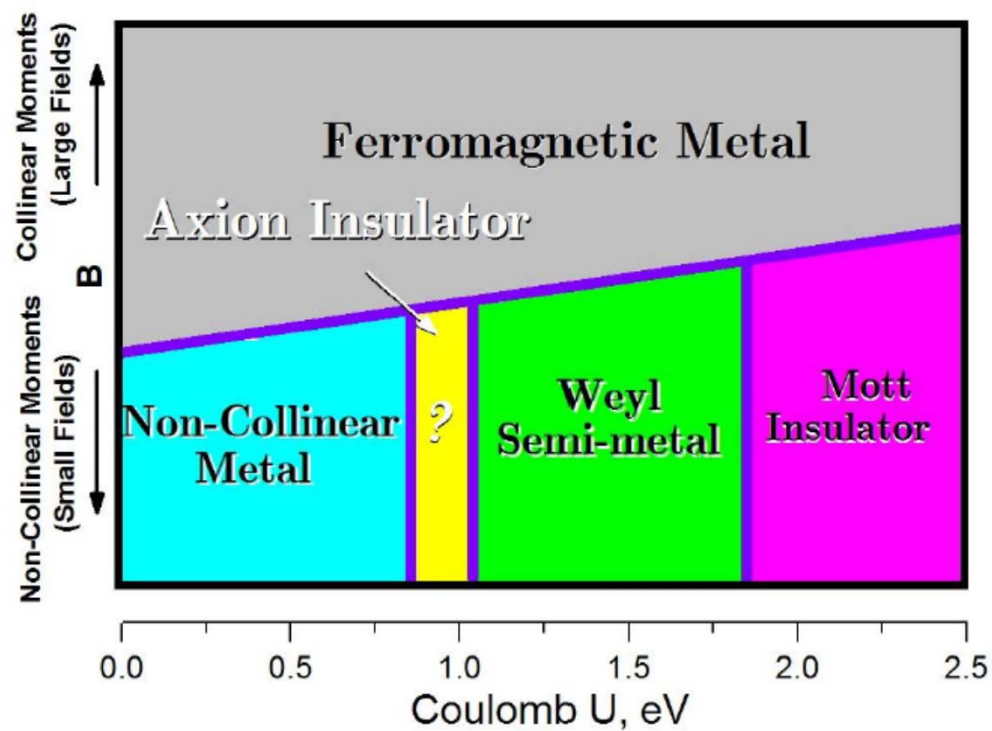
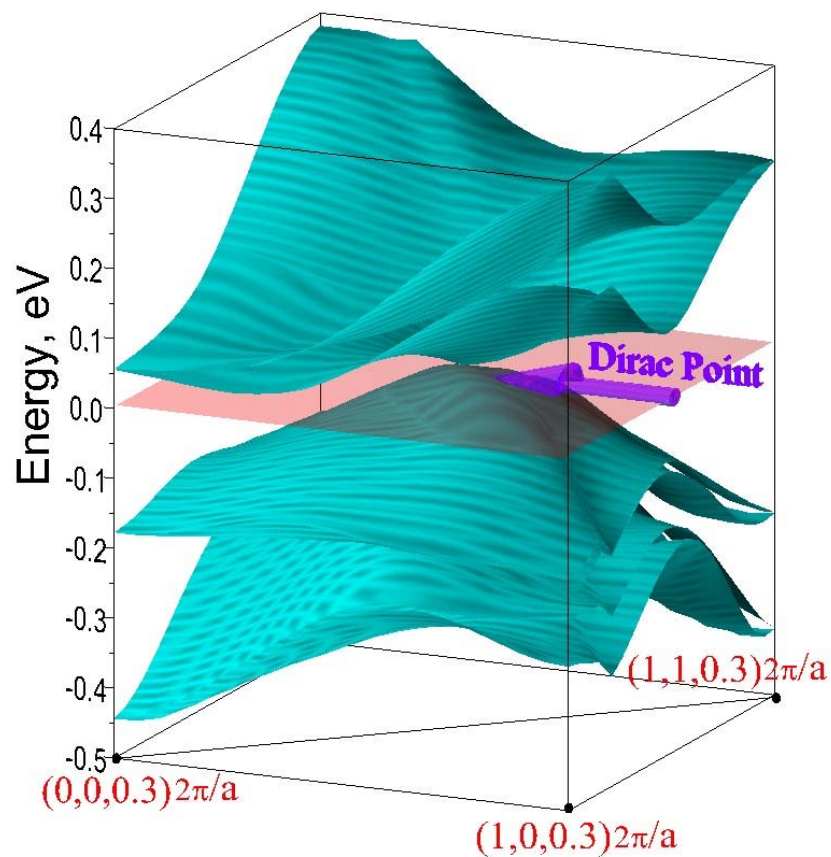
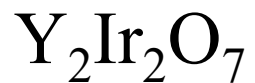
$U$



**+U calculation**

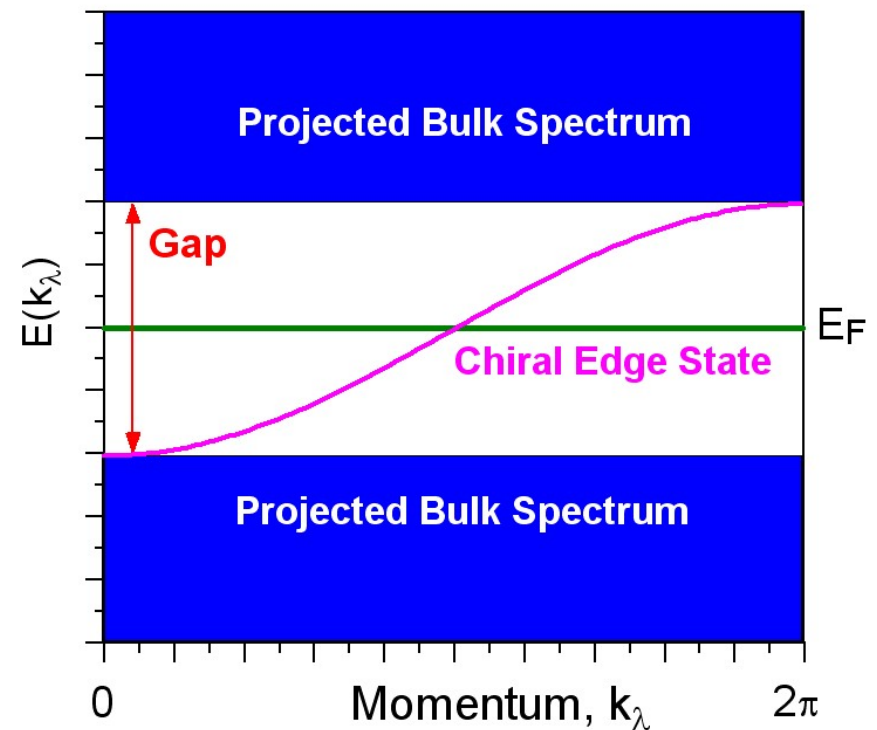
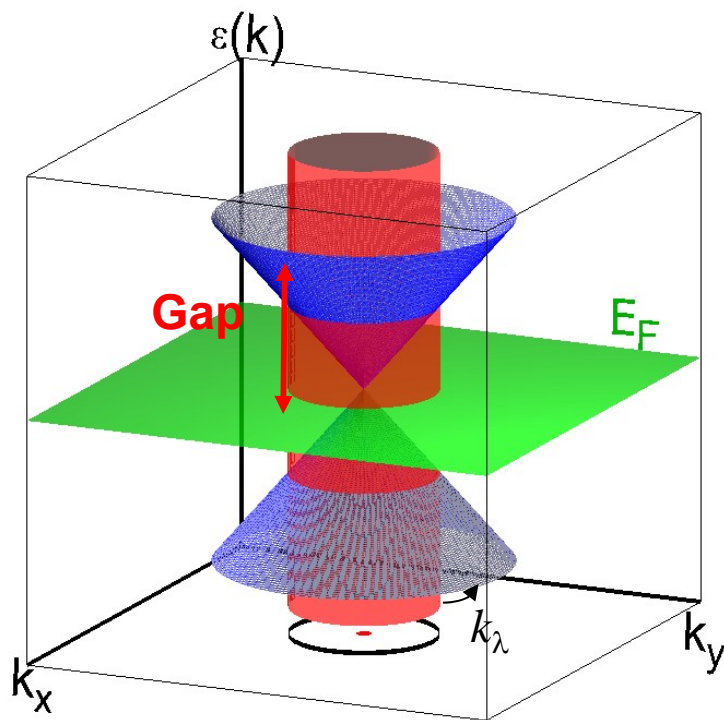
Wan et al., (2011)

# Weyl Semimetal



Wan, Turner, Vishwanath, Savrasov, Phys. Rev. B **83**, 205101 (2011)

# Fermi Arc (费米面是不连续的线段)



# Weyl Semimetal

## TaAs:

### Theory:

Weng et al., PRX (2015)

Huang et al., Nature Commum. (2015)

### Exp:

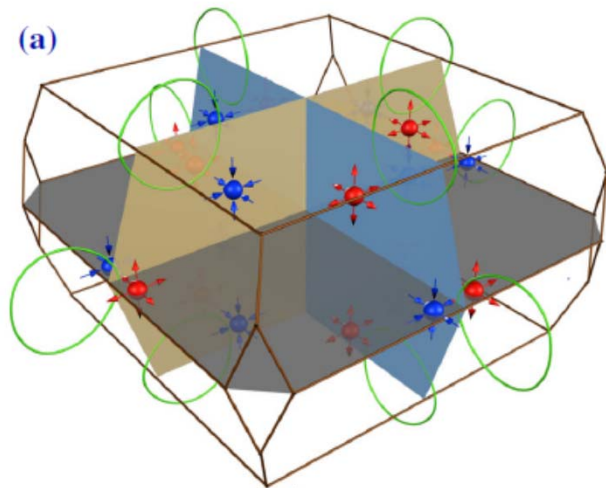
Lv et al., PRX (2015), Nature Phys. (2015);

Xu et al., Science (2015);

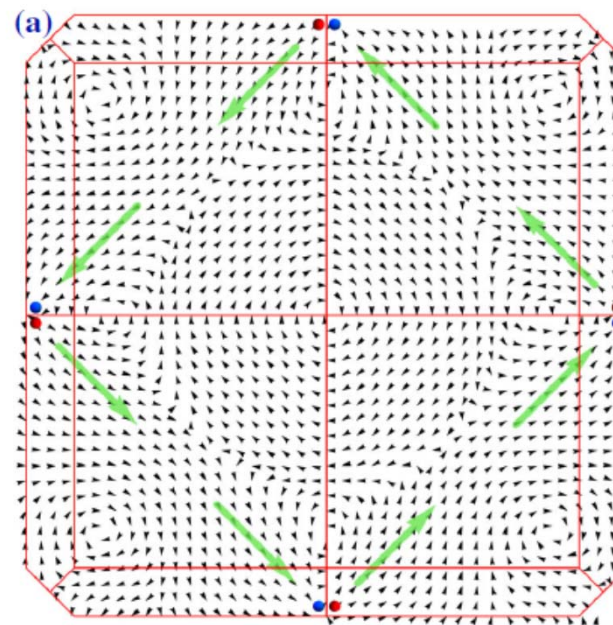
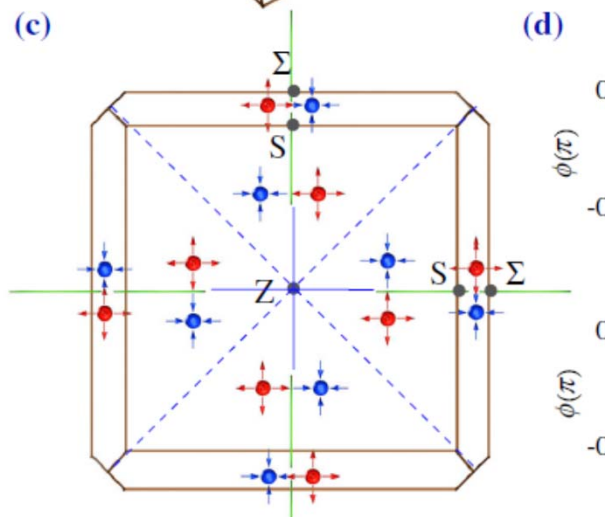
.....

# Weyl Semimetal Phase in Noncentrosymmetric Transition-Metal Monophosphides

Hongming Weng,<sup>1,2,\*</sup> Chen Fang,<sup>3</sup> Zhong Fang,<sup>1,2</sup> B. Andrei Bernevig,<sup>4</sup> and Xi Dai<sup>1,2</sup>



3D view of the nodal rings (in the absence of SOC) and Weyl points (with SOC) in the BZ. Once the SOC is turned on, the nodal rings are gapped and give rise to Weyl points off the mirror planes

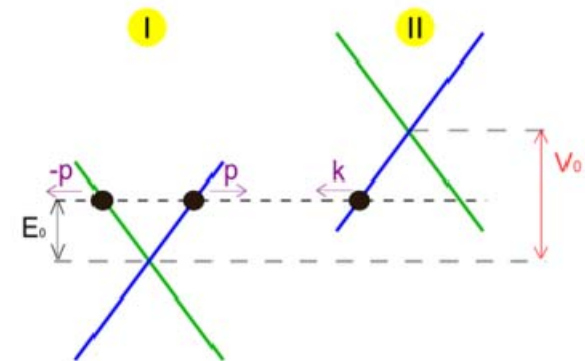


Berry curvature distribution



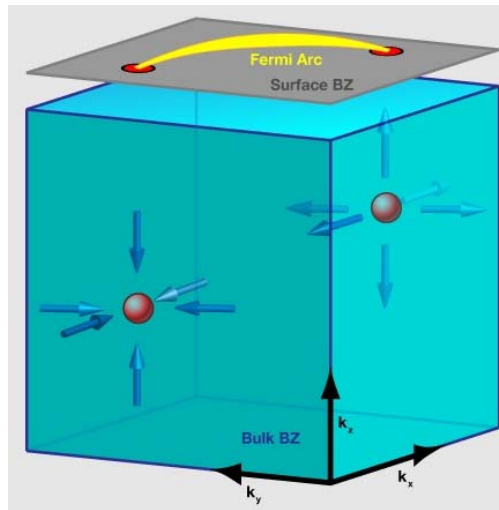
# Novel Properties of Quasiparticle

➤ **Klein tunneling** (Katsnelson et al., 2006)



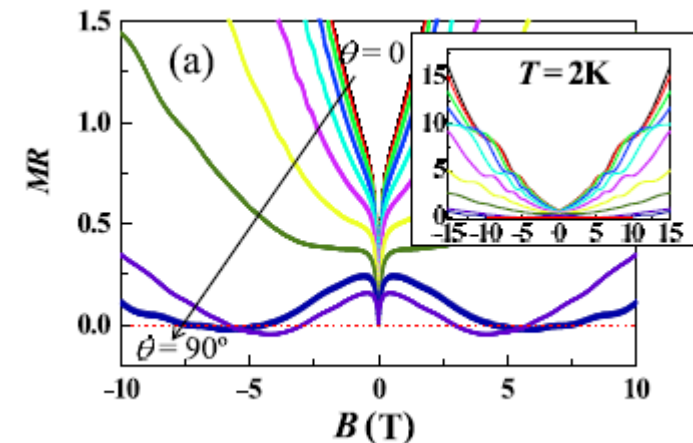
➤ **Fermi Arc**

(Wan, Turner, Vishwanath, Savrasov, (2011))



➤ **Chiral Anomaly**

(Aji (2011), Son & Spivak (2013))



# Topological (Crystalline) Insulators

➤ Z<sub>2</sub> topological insulator (*Kane&Mele PRL 2005*)

➤ Mirror Chern Insulator

*Ando and Fu, Annu. Rev. Condens. Matter Phys. 6, 361 (2015)*

➤ Hourglass Fermions

*Wang, Alexandradinata, Cava, Bernevig, Nature 532, 189 (2016).*

➤ Higher-order topological insulators

*Fang&Fu;Schindler et al., Langbehn et al., Song, Fang&Fang;Benalcazar et al., (2017)*

➤ Quantized electric multipole insulators

*Benalcazar, Bernevig, Hughes, Science 357, 61 (2017)*

➤ Nodal-chain metals

*Bzdušek, Wu, Rüegg, Sigrist & Soluyanov, Nature 538, 75 (2016).*

➤ Three-fold (or higher) band degeneracies Semimetal

*Bradlyn et al., Science (2016)*

# Semimetal

- Dirac Semimetal
- Weyl Semimetal
- Multi-degeneracy points
- Nodal-line semimetal
- Hopf-line Semimetal
- .....



# How to find Topo Mater

There are so much topological blablabla

Many topological invariants

➤ **Wave function?!!!**

➤ **Need calculate all** topological invariants for one material

**Thus find any new topo-mater is a big success**

# Crystal Symmetry

- Analyzing various crystallographic symmetries

*Fu & Kane (2007); parity  $\rightarrow$   $\mathbb{Z}_2$  Topological insulator*

*Fang, Gilbert & Bernevig, (2012); Slager et al., (2012);*

*Fang & Fu (2015); Fang & Fu (2017); Zhou et al., (2018).....*

## **Do not use wave-function**

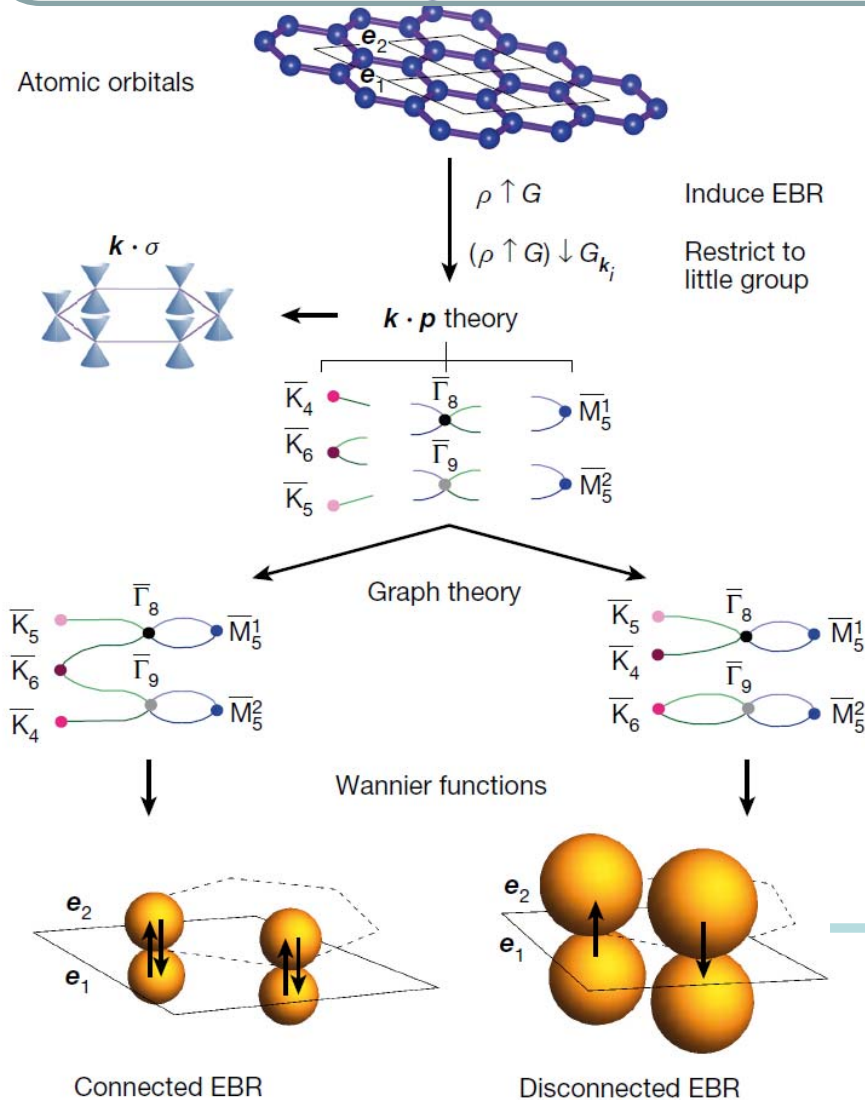
- Exploiting the mismatch between the real and momentum-space descriptions of the band structure

*Kruthoff et al., PRX (2017); Bradlyn et al., Nature (2017);*

*Po, Vishwanath & Watanabe, Nature Communications (2017)*

# Topological quantum chemistry

Barry Bradlyn<sup>1\*</sup>, L. Elcoro<sup>2\*</sup>, Jennifer Cano<sup>1\*</sup>, M. G. Vergniory<sup>3,4,5\*</sup>, Zhijun Wang<sup>6\*</sup>, C. Felser<sup>7</sup>, M. I. Aroyo<sup>2</sup> & B. Andrei Bernevig<sup>3,6,8,9</sup>



**Atomic limit**

**Representation + Compatibility**

**Graph theory and band structure**

Connected bands (EBR)  $\rightarrow$  Semimetal

Disconnected bands (EBR)  $\rightarrow$  topo insulator

**In total 10403 different EBRs**



Topological semimetal



Topological insulator

# Symmetry-based Indicators of Band Topology in the 230 Space Groups

Hoi Chun Po,<sup>1,2</sup> Ashvin Vishwanath,<sup>1,2,\*</sup> and Haruki Watanabe<sup>3</sup>

Band structure (has direct band gap at high symmetry points)

**Atomic limit**

$$\{\text{BS}\} \equiv \ker \mathcal{C} \cap \mathbb{Z}^D \simeq \mathbb{Z}^{d_{\text{BS}}} \quad \text{BS} = \sum_{i=1}^{d_{\text{BS}}} m_i \mathbf{b}_i$$

**Quotient Group**

**Provide no insight**

**商群**

**into how to find or**

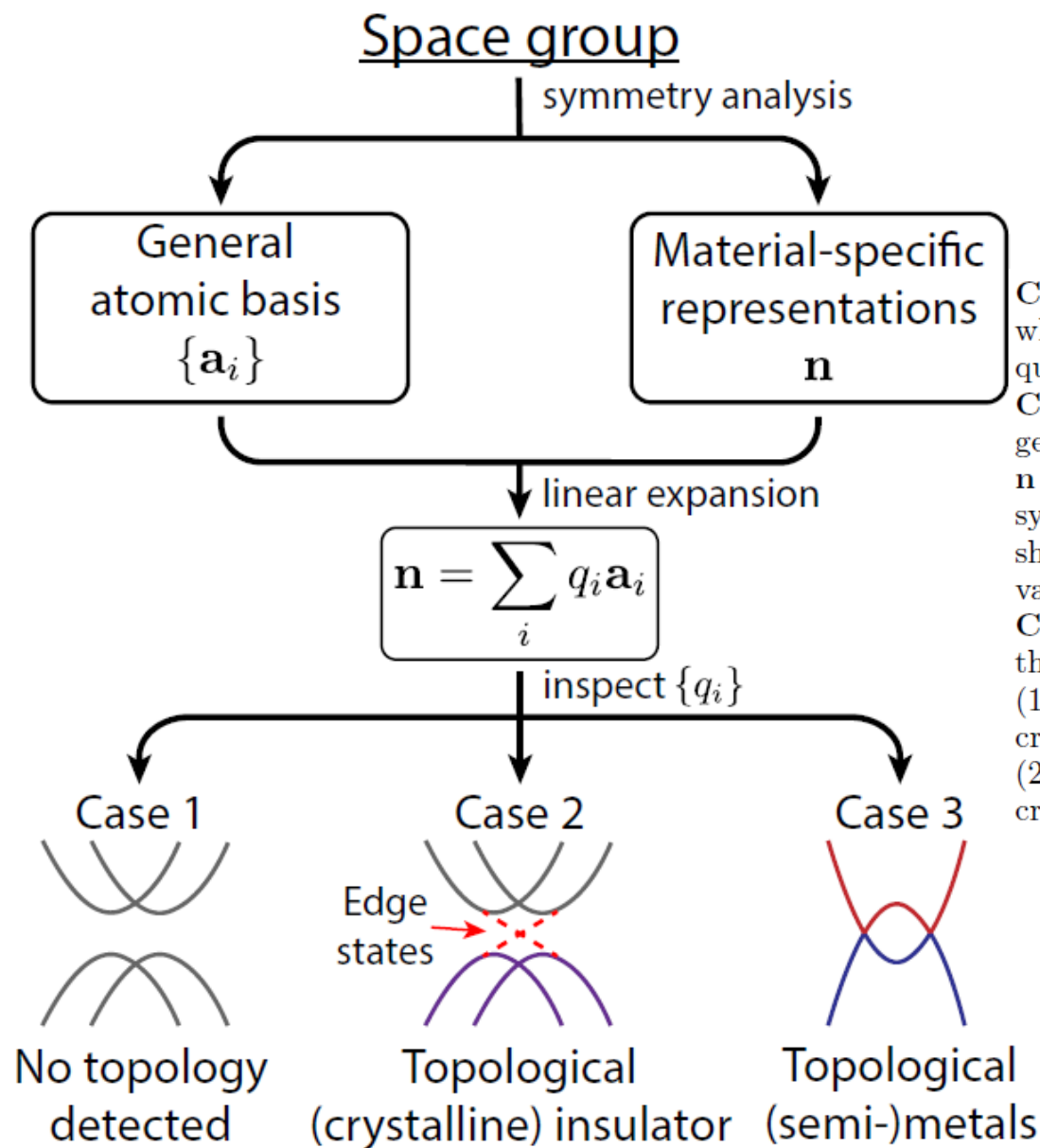
$$\{\text{AI}\} \simeq \mathbb{Z}^{d_{\text{AI}}} \equiv \left\{ \sum_{i=1}^{d_{\text{AI}}} m_i \mathbf{a}_i : m_i \in \mathbb{Z} \right\}$$

$$X_{\text{BS}} \equiv \frac{\{\text{BS}\}}{\{\text{AI}\}}$$

**engineer materials in  
any non-trivial class**

$X_{\text{BS}}$	Space groups
$\mathbb{Z}_2$	81, 82, 111, 112, 113, 114, 115, 116, 117, 118, 119, 120, 121, 122, 215, 216, 217, 218, 219, 220
$\mathbb{Z}_3$	188, 190
$\mathbb{Z}_4$	52, 56, 58, 60, 61, 62, 70, 88, 126, 130, 133, 135, 136, 137, 138, 141, 142, 163, 165, 167, 202, 203, 205, 222, 223, 227, 228, 230
$\mathbb{Z}_8$	128, 225, 226
$\mathbb{Z}_{12}$	176, 192, 193, 194
$\mathbb{Z}_2 \times \mathbb{Z}_4$	14, 15, 48, 50, 53, 54, 55, 57, 59, 63, 64, 66, 68, 71, 72, 73, 74, 84, 85, 86, 125, 129, 131, 132, 134, 147, 148, 162, 164, 166, 200, 201, 204, 206, 224
$\mathbb{Z}_2 \times \mathbb{Z}_8$	87, 124, 139, 140, 229
$\mathbb{Z}_3 \times \mathbb{Z}_3$	174, 187, 189
$\mathbb{Z}_4 \times \mathbb{Z}_8$	127, 221
$\mathbb{Z}_6 \times \mathbb{Z}_{12}$	175, 191
$\mathbb{Z}_2 \times \mathbb{Z}_2 \times \mathbb{Z}_4$	11, 12, 13, 49, 51, 65, 67, 69
$\mathbb{Z}_2 \times \mathbb{Z}_4 \times \mathbb{Z}_8$	83, 123
$\mathbb{Z}_2 \times \mathbb{Z}_2 \times \mathbb{Z}_2 \times \mathbb{Z}_4$	2, 10, 47

# Our Algorithm



$$\mathbf{n} = (\nu, n_{\mathbf{k}_1}^1, \dots, n_{\mathbf{k}_1}^{\alpha_1}, n_{\mathbf{k}_2}^1, \dots, n_{\mathbf{k}_2}^{\alpha_2}, \dots, n_{\mathbf{k}_N}^1, \dots, n_{\mathbf{k}_N}^{\alpha_N})$$

230 space group

Go beyond one-by-one

Case 1: the expansion coefficients  $q_i$ 's are all integers, which indicates that  $\mathbf{n}$  is an AI, thus the material is equivalent to an atomic insulator;

Case 2: the expansion coefficients  $q_i$ 's are not all integers, but all  $q_i C_i$ 's are integers. This indicates that the  $\mathbf{n}$  is a BS and the material is equivalent to a BI, but this system must be topological since some remainder(s) should be nonzero as shown in Eq. 3, i.e. the SI is non-vanishing;

Case 3: Not only the expansion coefficients  $q_i$ 's but also the  $q_i C_i$ 's are not all integers:

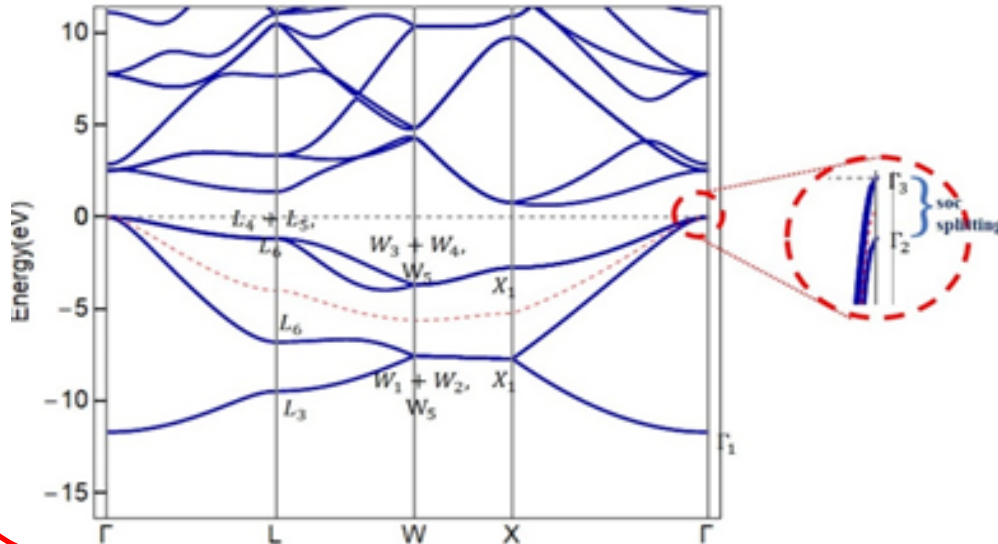
(1) some  $n_{\mathbf{k}}^\alpha$  is non-integer which indicates that band crossing happens at this  $k$  point;

(2) all the  $n_{\mathbf{k}}^{\alpha'}$ 's are integers, then there must be band crossing in high symmetry line or plane.

*Tang, Po, Vishwanath, Wan\**  
*15, 470 Nature Physics (2019)*

# Band Structure

**Band structure of Si  
(SG227)**



**Topological** is Global

**What is detail:**  
If nothing pass the Fermi level, then all is detail!

**BZ**

《物理》 48, 341 (2019)

高对称点	坐标	Irreps
$\Gamma$	(0, 0, 0)	$G_{96}^8$ : 4(2), 5(2), 8(4), 12(2), 13(2), 16(4)
$X$	(0, 1, 0)	$G_{64}^2$ : 19(4)
$L$	(1/2, 1/2, 1/2)	$G_{24}^3$ : 3(1), 4(1), 6(2), 9(1), 10(1), 12(2)
$W$	(1/2, 1, 0)	$G_{64}^3$ : 13(1), 14(1), 15(1), 16(1), 20(2)

Bands can cross when they carry different symmetry labels

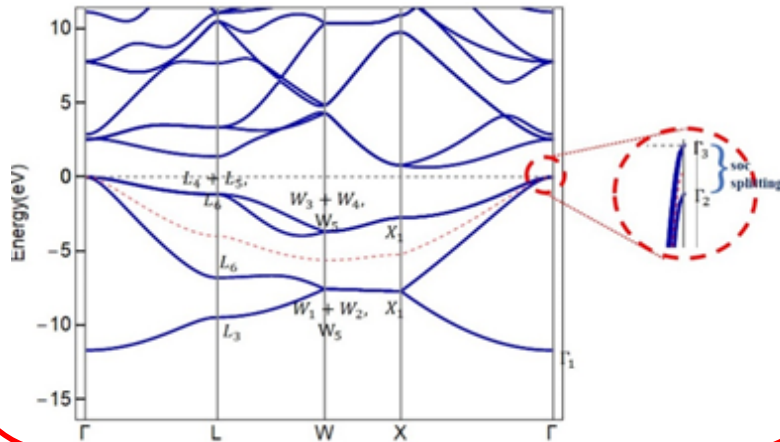
Dimensions of the irreps determine how the bands are “stuck”

**High symmetry point, High symmetry line, irreps, .....**



# Band Structure

Band structure of Si



What is detail:

If nothing pass the Fermi level, then all is detail!

$$\mathbf{n}_{\text{BS}} = \left( \nu, n_{k_1}^1, n_{k_1}^2, \dots, n_{k_1}^{r_1}, n_{k_2}^1, n_{k_2}^2, \dots, n_{k_2}^{r_2}, \dots, n_{k_N}^1, n_{k_N}^2, \dots, n_{k_N}^{r_N} \right).$$

高对称点	坐标	Irreps
$\Gamma$	(0, 0, 0)	$G_{96}^8$ : 4(2), 5(2), 8(4), 12(2), 13(2), 16(4)
X	(0, 1, 0)	$G_{64}^2$ : 19(4)
L	(1/2, 1/2, 1/2)	$G_{24}^3$ : 3(1), 4(1), 6(2), 9(1), 10(1), 12(2)
W	(1/2, 1, 0)	$G_{64}^3$ : 13(1), 14(1), 15(1), 16(1), 20(2)

$$\mathbf{n}_{\text{BS}} = (8, 1, 1, 1, 0, 0, 0, 2, 0, 0, 1, 1, 1, 2, 1, 1, 1, 1, 2).$$

$$\mathbf{n}_{\text{BS}, 1} = (4, 0, 0, 1, 0, 0, 0, 1, 0, 0, 0, 1, 1, 1, 0, 0, 1, 1, 1).$$

$$\mathbf{n}_{\text{BS}, 2} = (4, 1, 1, 0, 0, 0, 0, 1, 0, 0, 1, 0, 0, 1, 1, 1, 0, 0, 1).$$

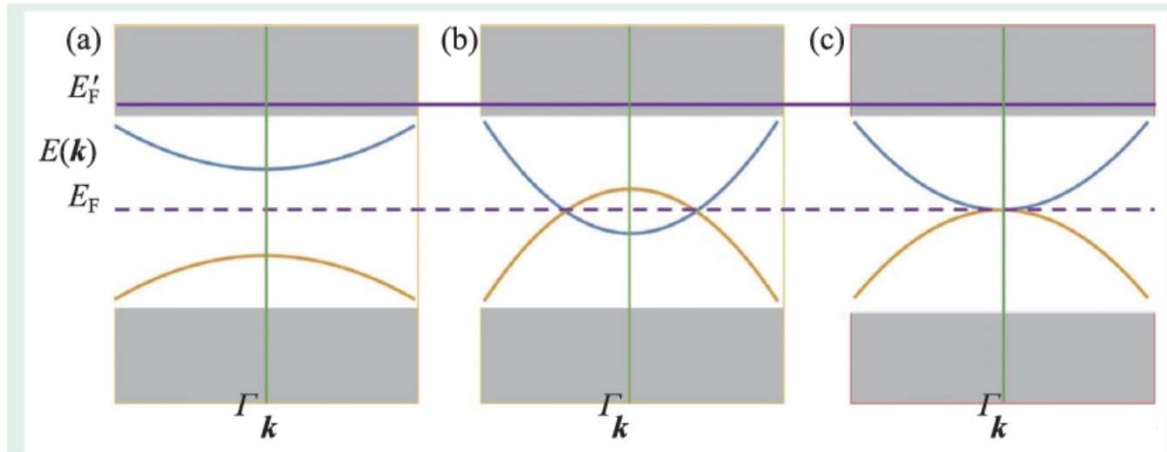
$$\mathbf{n}_{\text{BS}} = \mathbf{n}_{\text{BS}, 1} + \mathbf{n}_{\text{BS}, 2}$$

➤ Labels become simple counting!

➤ Gap conditions above and below ensure counting is well defined



# Atomic limit



- Unit cell has one atom
- One  $s$  orbit at this atom

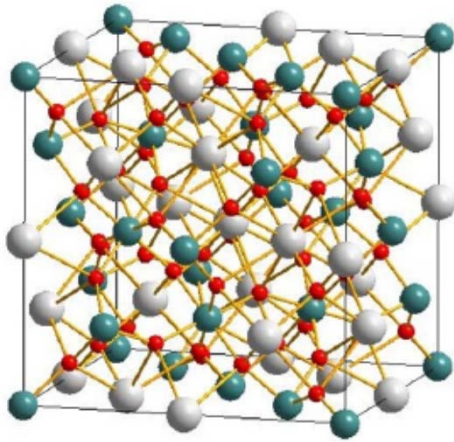
230 Space group

- Wyckoff Positions
- Site symmetry

## Wyckoff Positions of Group 2 ( P -1)

Multiplicity	Wyckoff letter	Site symmetry	Coordinates
2	i	1	$(x, y, z) (-x, -y, -z)$
1	h	-1	$(1/2, 1/2, 1/2)$
1	g	-1	$(0, 1/2, 1/2)$
1	f	-1	$(1/2, 0, 1/2)$
1	e	-1	$(1/2, 1/2, 0)$
1	d	-1	$(1/2, 0, 0)$
1	c	-1	$(0, 1/2, 0)$
1	b	-1	$(0, 0, 1/2)$
1	a	-1	$(0, 0, 0)$

## Wyckoff Positions of Group 227 ( F d -3 m ) [origin choice]



Multiplicity	Wyckoff letter	Site symmetry	Coordinates						
			( 0,0,0)+ ( 0,1/2,1/2)+ ( 1/2,0,1/2)+ ( 1/2,1/2,0)+						
192	i	1	(x, y, z)	( - x, - y + 1/2, z + 1/2 )	( - x + 1/2, y + 1/2, - z )				
			(z, x, y)	( z + 1/2, - x, - y + 1/2 )	( - z, - x + 1/2, y + 1/2 )				
			(y, z, x)	( - y + 1/2, z + 1/2, - x )	( y + 1/2, - z, - x + 1/2 )				
			(y + 3/4, x + 1/4, - z + 3/4 )	( - y + 1/4, - x + 1/4, - z + 1/4 )	( y + 1/4, - x + 3/4, z + 3/4 )				
			(x + 3/4, z + 1/4, - y + 3/4 )	( - x + 3/4, z + 3/4, y + 1/4 )	( - x + 1/4, - z + 1/4, - y + 3/4 )				
			(z + 3/4, y + 1/4, - x + 3/4 )	( z + 1/4, - y + 3/4, x + 3/4 )	( - z + 3/4, y + 3/4, x + 1/4 )				
			( - x + 1/4, - y + 1/4, - z + 1/4 )	( x + 1/4, y + 3/4, - z + 3/4 )	( x + 3/4, - y + 3/4, z + 1/4 )				
			( - z + 1/4, - x + 1/4, - y + 1/4 )	( - z + 3/4, x + 1/4, y + 3/4 )	( z + 1/4, x + 3/4, - y + 3/4 )				
			( - y + 1/4, - z + 1/4, - x + 1/4 )	( y + 3/4, - z + 3/4, x + 1/4 )	( - y + 3/4, z + 1/4, x + 3/4 )				
			( - y + 1/2, - x, z + 1/2 )	( y, x, z )	( - y, x + 1/2, - z + 1/2 )				
			( - x + 1/2, - z, y + 1/2 )	( x + 1/2, - z + 1/2, - y )	( x, z, y )				
			( - z + 1/2, - y, x + 1/2 )	( - z, y + 1/2, - x + 1/2 )	( z + 1/2, - y + 1/2, - x )				
			96	h	..2	(1/8, y, - y + 1/4 )	( 7/8, - y + 1/2, - y + 3/4 )	( 3/8, y + 1/2, y + 3/4 )	( 5/8, - y, y + 1/4 )
						( - y + 1/4, 1/8, y )	( - y + 3/4, 7/8, - y + 1/2 )	( y + 3/4, 3/8, y + 1/2 )	( y + 1/4, 5/8, - y )
(y, - y + 1/4, 1/8 )	( - y + 1/2, - y + 3/4, 7/8 )	( y + 1/2, y + 3/4, 3/8 )				( - y, y + 1/4, 5/8 )			
(1/8, - y + 1/4, y )	( 3/8, y + 3/4, y + 1/2 )	( 7/8, - y + 3/4, - y + 1/2 )				( 5/8, y + 1/4, - y )			
(y, 1/8, - y + 1/4 )	( y + 1/2, 3/8, y + 3/4 )	( - y + 1/2, 7/8, - y + 3/4 )				( - y, 5/8, y + 1/4 )			
( - y + 1/4, y, 1/8 )	( y + 3/4, y + 1/2, 3/8 )	( - y + 3/4, - y + 1/2, 7/8 )				( y + 1/4, - y, 5/8 )			
96	g	..m				(x, x, z)	( - x, - x + 1/2, z + 1/2 )	( - x + 1/2, x + 1/2, - z )	
						(z, x, x)	( z + 1/2, - x, - x + 1/2 )	( - z, - x + 1/2, x + 1/2 )	
						(x, z, x)	( - x + 1/2, z + 1/2, - x )	( x + 1/2, - z, - x + 1/2 )	
						(x + 3/4, x + 1/4, - z + 3/4 )	( - x + 1/4, - x + 1/4, - z + 1/4 )	( x + 1/4, - x + 3/4, z + 3/4 )	

**Pyrochlore structure:** Space Group 227

Y → 16d

Ir → 16c

O1 → 8b

O2 → 48f

# Calculation of Atomic Insulator Basis

Taking SG2 as an example

*Step 1:*

*Obtaining HSPs.*

$SG2's$ HSP	$\Gamma$	$X$	$Y$	$Z$	$U$	$T$	$S$	$R$
coordinate	(0,0,0)	( $\frac{1}{2}$ , 0, 0)	(0, $\frac{1}{2}$ , 0)	(0, 0, $\frac{1}{2}$ )	( $\frac{1}{2}$ , $\frac{1}{2}$ , 0)	(0, $\frac{1}{2}$ , $\frac{1}{2}$ )	( $\frac{1}{2}$ , 0, $\frac{1}{2}$ )	( $\frac{1}{2}$ , $\frac{1}{2}$ , $\frac{1}{2}$ )

*Step 2: Give all the Wyckoff positions.*

2  $\triangleleft$  8  $\equiv$  1  $\blacksquare$  17

$SG2's$ Wyckoff position	site-group	Wyckoff orbits
2i	$C_1$	$(x, y, z), (-x, -y, -z)$
1h	$C_i$	$(1/2, 1/2, 1/2)$
1g	$C_i$	$(0, 1/2, 1/2)$
1f	$C_i$	$(1/2, 0, 1/2)$
1e	$C_i$	$(1/2, 1/2, 0)$
1d	$C_i$	$(1/2, 0, 0)$
1c	$C_i$	$(0, 1/2, 0)$
1b	$C_i$	$(0, 0, 1/2)$
1a	$C_i$	$(0, 0, 0)$

TABLE VII. The nine Wyckoff positions for  $SG2$ .

# Calculation of Atomic Insulator Basis

$SG2$	$n_{2i}^1$	$n_{1h}^1$	$n_{1h}^2$	$n_{1g}^1$	$n_{1g}^2$	$n_{1f}^1$	$n_{1f}^2$	$n_{1e}^1$	$n_{1e}^2$	$n_{1d}^1$	$n_{1d}^2$	$n_{1c}^1$	$n_{1c}^2$	$n_{1b}^1$	$n_{1b}^2$	$n_{1a}^1$	$n_{1a}^2$
$\nu$	4	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2
$\Gamma_1^1$	1	1	0	1	0	1	0	1	0	1	0	1	0	1	0	1	0
$\Gamma_2^1$	1	0	1	0	1	0	1	0	1	0	1	0	1	0	1	0	1
$X_1^1$	1	1	0	0	1	1	0	0	1	0	1	0	1	1	0	1	0
$X_2^1$	1	0	1	1	0	0	1	1	0	1	0	1	0	0	1	0	1
$Y_1^1$	1	1	0	0	1	0	1	1	0	0	1	1	0	0	1	1	0
$Y_2^1$	1	0	1	1	0	1	0	0	1	1	0	0	1	1	0	0	1
$Z_1^1$	1	1	0	0	1	0	1	0	1	1	0	1	0	1	0	0	1
$Z_2^1$	1	0	1	1	0	1	0	1	0	0	1	0	1	0	1	1	0
$U_1^1$	1	1	0	1	0	0	1	0	1	1	0	0	1	0	1	1	0
$U_2^1$	1	0	1	0	1	1	0	1	0	0	1	1	0	1	0	0	1
$T_1^1$	1	1	0	1	0	1	0	0	1	0	1	1	0	0	1	0	1
$T_2^1$	1	0	1	0	1	0	1	1	0	1	0	0	1	1	0	1	0
$S_1^1$	1	1	0	1	0	0	1	1	0	0	1	0	1	1	0	0	1
$S_2^1$	1	0	1	0	1	1	0	0	1	1	0	1	0	0	1	1	0
$R_1^1$	1	1	0	0	1	1	0	1	0	1	0	0	1	0	1	0	1
$R_2^1$	1	0	1	1	0	0	1	0	1	0	1	1	0	1	0	1	0

TABLE X. The 17 AI vectors for  $SG2$ .

$$\mathbf{n} = (\nu, n_{\mathbf{k}_1}^1, n_{\mathbf{k}_1}^2, \dots, n_{\mathbf{k}_1}^{\alpha_1}, \dots, n_{\mathbf{k}_1}^{r_1}, n_{\mathbf{k}_2}^1, n_{\mathbf{k}_2}^2, \dots, n_{\mathbf{k}_2}^{\alpha_2}, \dots, n_{\mathbf{k}_2}^{r_2}, \dots, n_{\mathbf{k}_i}^1, \dots, n_{\mathbf{k}_i}^{\alpha_i}, \dots, n_{\mathbf{k}_i}^{r_i}, \dots, n_{\mathbf{k}_N}^1, \dots, n_{\mathbf{k}_N}^{\alpha_N}, \dots, n_{\mathbf{k}_N}^{r_N}).$$

The BS can be represented by an integer-valued vector,  $\mathbf{n} =$

$$(\nu, n_{\mathbf{k}_1}^1, n_{\mathbf{k}_1}^2, \dots, n_{\mathbf{k}_1}^{\alpha_1}, \dots, n_{\mathbf{k}_1}^{r_1}, n_{\mathbf{k}_2}^1, n_{\mathbf{k}_2}^2, \dots, n_{\mathbf{k}_2}^{\alpha_2}, \dots, n_{\mathbf{k}_2}^{r_2}, \dots, n_{\mathbf{k}_i}^1, \dots, n_{\mathbf{k}_i}^{\alpha_i}, \dots, n_{\mathbf{k}_i}^{r_i}, \dots, n_{\mathbf{k}_N}^1, \dots, n_{\mathbf{k}_N}^{\alpha_N}, \dots, n_{\mathbf{k}_N}^{r_N}).$$

# Calculation of Atomic Insulator Basis

We choose our basis such that the values of  $C_i$  are maximized: **Smith decomposition**.

(a) The 9 AI basis vectors for  $SG2$ . Here  $\nu$  is the number of the bands. It is also called the filling number. Starting from the 3rd row, we give the number  $n_k^\alpha$  in order. We omit the notation  $n$  for clarity: The first column of these rows gives the information of the HSP and its irrep completely.

$$\mathbf{Z}_2 \quad \triangleleft \mathbf{Z}_2 \quad \triangleleft \mathbf{Z}_2 \quad \triangleleft \mathbf{Z}_4$$

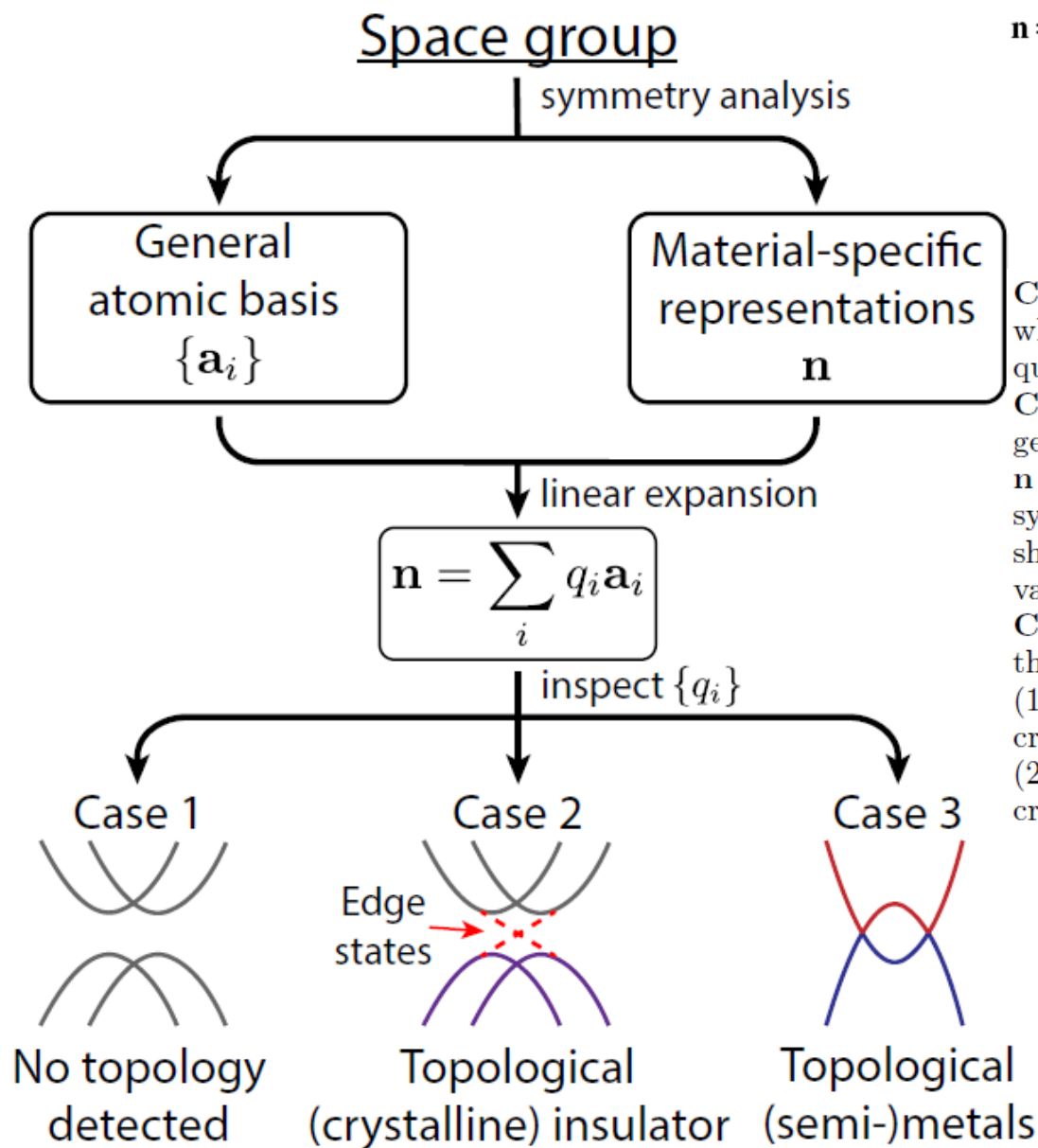
$$\text{BS} = \sum_{i=1}^{d_{\text{BS}}} m_i \mathbf{b}_i,$$

**Common factors**

$$d_{\text{AI}} = d_{\text{BS}} \quad \text{Po et al., NC (2017)}$$

$SG2$	$a_1$	$a_2$	$a_3$	$a_4$	$a_5$	$a_6$	$a_7$	$a_8$	$a_9$
$\nu$	4	2	2	2	2	4	4	4	8
$\Gamma_1^1$	1	1	1	1	1	2	2	2	4
$\Gamma_2^1$	1	0	0	0	0	0	0	0	0
$X_1^1$	1	1	0	1	0	2	2	0	4
$X_2^1$	1	0	1	0	1	0	0	2	0
$Y_1^1$	1	1	0	0	1	2	0	2	4
$Y_2^1$	1	0	1	1	0	0	2	0	0
$Z_1^1$	1	1	0	0	0	0	0	0	0
$Z_2^1$	1	0	1	1	1	2	2	2	4
$U_1^1$	1	1	1	0	0	0	2	2	4
$U_2^1$	1	0	0	1	1	2	0	0	0
$T_1^1$	1	1	1	1	0	2	2	2	4
$T_2^1$	1	0	0	0	1	0	0	0	0
$S_1^1$	1	1	1	0	1	2	2	2	4
$S_2^1$	1	0	0	1	0	0	0	0	0
$R_1^1$	1	1	0	1	1	2	2	2	4
$R_2^1$	1	0	1	0	0	0	0	0	0

# Our Algorithm



$$\mathbf{n} = (\nu, n_{\mathbf{k}_1^1}, \dots, n_{\mathbf{k}_1^{\alpha_1}}, n_{\mathbf{k}_2^1}, \dots, n_{\mathbf{k}_2^{\alpha_2}}, \dots, n_{\mathbf{k}_N^1}, \dots, n_{\mathbf{k}_N^{\alpha_N}})$$

**230 space group**  
**Go beyond one-by-one**

Case 1: the expansion coefficients  $q_i$ 's are all integers, which indicates that  $\mathbf{n}$  is an AI, thus the material is equivalent to an atomic insulator;

Case 2: the expansion coefficients  $q_i$ 's are not all integers, but all  $q_i C_i$ 's are integers. This indicates that the  $\mathbf{n}$  is a BS and the material is equivalent to a BI, but this system must be topological since some remainder(s) should be nonzero as shown in Eq. 3, i.e. the SI is non-vanishing;

Case 3: Not only the expansion coefficients  $q_i$ 's but also the  $q_i C_i$ 's are not all integers:

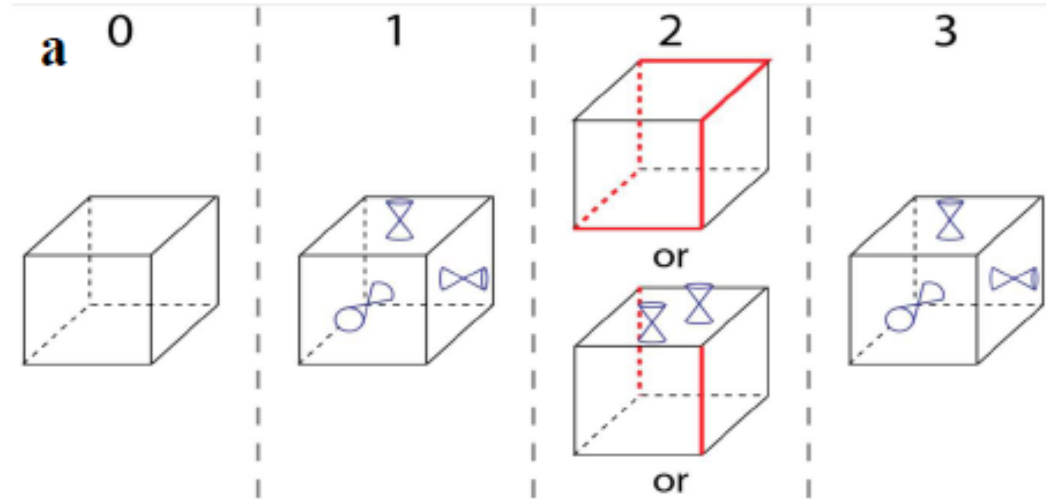
(1) some  $n_{\mathbf{k}}^\alpha$  is non-integer which indicates that band crossing happens at this  $k$  point;

(2) all the  $n_{\mathbf{k}}^{\alpha'}$ s are integers, then there must be band crossing in high symmetry line or plane.

*Tang, Po, Vishwanath, Wan\**  
*15, 470 Nature Physics (2019)*



# $\mathbb{Z}_4$ TCI



**2 of  $\mathbb{Z}_4$**

TABLE I. We focus on the following space groups ( $\mathcal{SG}$ s), in which a strong topological insulators generate a  $\mathbb{Z}_4$  subgroup in the group of symmetry indicators,  $X_{\text{BS}}$ . The entry  $2 \in \mathbb{Z}_4$  corresponds to various kinds of topological crystalline insulators, and the predicted materials candidates for such phases are tabulated.

$X_{\text{BS}}$	$\mathbb{Z}_2^3 \times \mathbb{Z}_4$	$\mathbb{Z}_2^2 \times \mathbb{Z}_4$	$\mathbb{Z}_2 \times \mathbb{Z}_4$	$\mathbb{Z}_4$
$\mathcal{SG}$	<b>2</b>	<b>11,12</b>	<b>166</b>	<b>61, 136, 227</b>
Materials	$\text{Ag}_2\text{F}_5$	$\beta\text{-MoTe}_2, \text{BiBr}$	A7-P	c-TiS <sub>2</sub>



# Topological Hinge states in $\beta$ -MoTe<sub>2</sub>

(SG) 11 ( $P2_1/m$ ).

$\alpha$ -phase:

**$\beta$ -phase: room temperature**

Monolayer—QSH (*Qian, Liu, Fu & Li, Science 2014*)

$\gamma$ -phase:

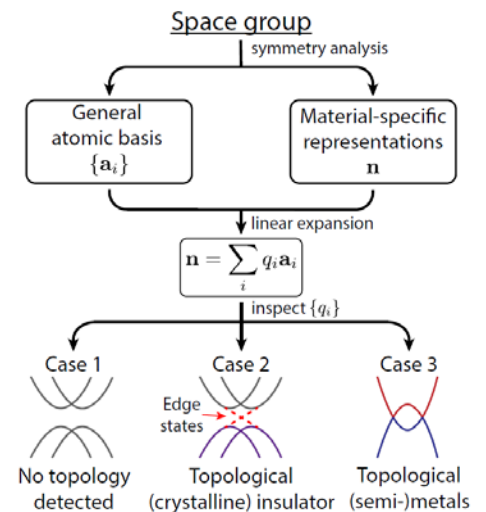
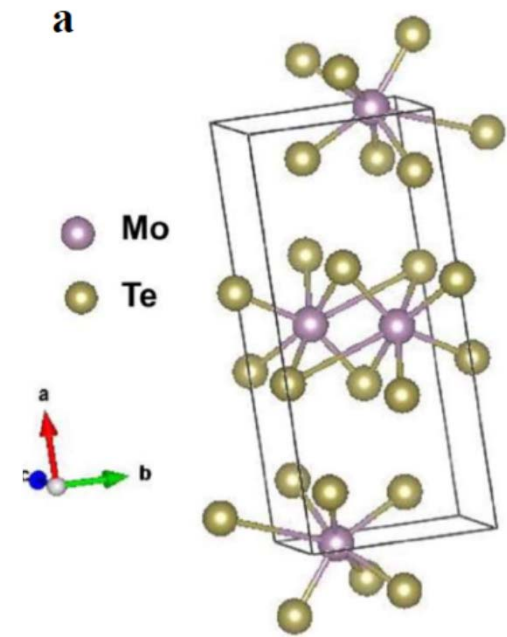
Type-II Weyl Semimetal:

*Deng et al., Nature Phys. (2016)*

*Tamai et al., PRX (2016)*

*Huang et al., Nature Mater. (2016)*

*Jiang et al., Nature Commun. (2016)*



# DFT results of $\beta$ -MoTe<sub>2</sub>

- 2 inequivalent Mo's and 4 in-equivalent Te's
  - all occupy 2e Wyckoff positions.
  - total 12 atoms (56 valence electrons) primitive unit cell
- v=56                  SG11**

TABLE I. For  $SG11$ , the HSPs are given by the labels  $\Gamma, B, \dots$  in order. For the labeling of the irreps of  $\mathcal{G}(\mathbf{k}_i)$ , we use  $(j, m)$  where  $j$  means the  $j$ th irrep and  $m$  denotes the dimension of the corresponding irrep. They are all listed in Ref. [2]. We use the same order of the irrep as Ref. [2]. The red color means that due to  $\mathcal{T}$ , the irrep must occur with its  $\mathcal{T}$  pair (belonging to the same irrep) simultaneously. Thus  $\mathcal{T}$  requires that the red colored irreps must happen even times. So it is necessary to divide them by 2 [1] to obtain the physical common factors.

HSP	$\Gamma$				$B$				$Y$				$Z$	$C$	$D$	$A$				$E$
irrep	(1,1)	(2,1)	(3,1)	(4,1)	(1,1)	(2,1)	(3,1)	(4,1)	(1,1)	(2,1)	(3,1)	(4,1)	(1,2)	(1,2)	(1,2)	(1,1)	(2,1)	(3,1)	(4,1)	(1,2)
$n_{\mathbf{k}_i}^{\alpha_i}$	16	16	12	12	14	14	14	14	14	14	14	14	14	14	14	14	14	14	14	14

$$\mathbf{n} = (\nu, n_{\mathbf{k}_1}^1, n_{\mathbf{k}_1}^2, \dots, n_{\mathbf{k}_1}^{\alpha_1}, \dots, n_{\mathbf{k}_2}^1, n_{\mathbf{k}_2}^2, \dots, n_{\mathbf{k}_2}^{\alpha_2}, \dots, n_{\mathbf{k}_N}^1, n_{\mathbf{k}_N}^2, \dots, n_{\mathbf{k}_N}^{\alpha_N}, \dots),$$

**DFT**



$$\mathbf{n} = (56, 16, 16, 12, 12, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14)$$

# Expansion of $\beta\text{-MoTe}_2$

(b) The 5 AI basis vectors for  $SG11$ .

$SG11$	$\mathbf{a}_1$	$\mathbf{a}_2$	$\mathbf{a}_3$	$\mathbf{a}_4$	$\mathbf{a}_5$
$\nu$	4	4	0	0	0
$\Gamma_1^1$	1	2	0	0	0
$\Gamma_2^1$	1	2	0	0	0
$\Gamma_3^1$	1	0	0	0	0
$\Gamma_4^1$	1	0	0	0	0
$B_1^1$	1	0	0	2	0
$B_2^1$	1	0	0	2	0
$B_3^1$	1	2	0	-2	0
$B_4^1$	1	2	0	-2	0
$Y_1^1$	1	0	2	0	0
$Y_2^1$	1	0	2	0	0
$Y_3^1$	1	2	-2	0	0
$Y_4^2$	1	2	-2	0	0
$Z_1^2$	1	1	0	0	0
$C_1^2$	1	1	0	0	0
$D_1^2$	1	1	0	0	0
$A_1^1$	1	2	-2	-2	4
$A_2^1$	1	2	-2	-2	4
$A_3^1$	1	0	2	2	-4
$A_4^1$	1	0	2	2	-4
$E_1^2$	1	1	0	0	0

$$C_1 = C_2 = 1, C_3 = C_4 = 2, \text{ and } C_5 = 4$$

$$\mathbb{Z}_2 \times \mathbb{Z}_2 \times \mathbb{Z}_4$$

$$\mathbf{n} = (56, 16, 16, 12, 12, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14)$$

$$= 12\mathbf{a}_1 + 2\mathbf{a}_2 + \mathbf{a}_3 + \mathbf{a}_4 + \frac{1}{2}\mathbf{a}_5,$$

$$(q_1, q_2, q_3, q_4, q_5) = (12, 2, 1, 1, \frac{1}{2})$$

$$\text{SI } (0, 0, 2)$$

$$2 \in \mathbb{Z}_4$$

**2 of  $\mathbb{Z}_4$**

# Topological feature of $\beta$ -MoTe<sub>2</sub>

$$\kappa_1 = \sum_{\mathbf{k} \in \text{TRIM}} (n_{\mathbf{k}}^+ - n_{\mathbf{k}}^-)/2 \pmod{4}$$

Z. Song, T. Zhang, Z. Fang, and C. Fang, ArXiv e-prints (2017), [arXiv:1711.11049](#).

E. Khalaf, H. C. Po, A. Vishwanath, and H. Watanabe, ArXiv e-prints (2017), [arXiv:1711.11589](#).

$\mathbf{k} \in \text{TRIM}$	$\Gamma$	$X$	$Y$	$Z$	$U$	$T$	$S$	$R$
$n_{\mathbf{k}}^+$	16	14	14	14	14	14	14	14
$n_{\mathbf{k}}^-$	12	14	14	14	14	14	14	14

$n_{\mathbf{k}}^{\pm}$  is the number of the occupied even/odd Kramers pairs

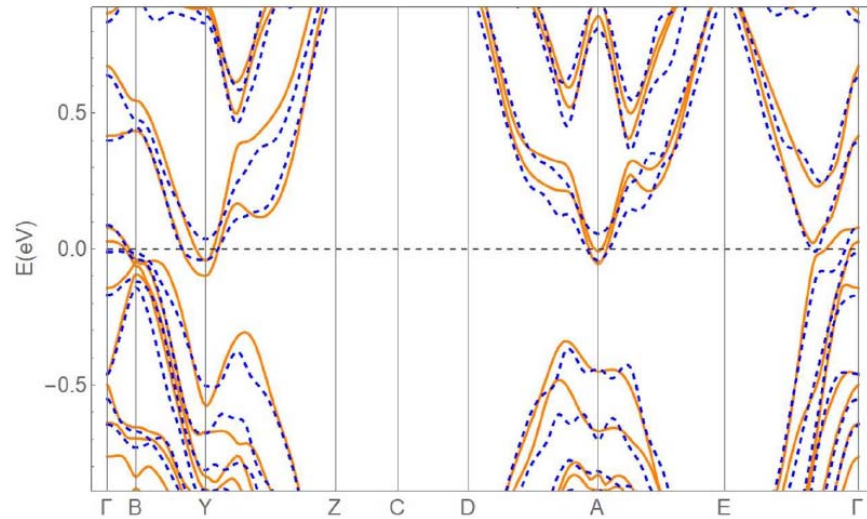
For this compound, 2 of  $Z_4 \rightarrow$  mirror Chern/hinge states

**WIEN2K**  $\rightarrow$  mirror Chern number for  $k_z = 0$  and  $k_z = \pi/c$

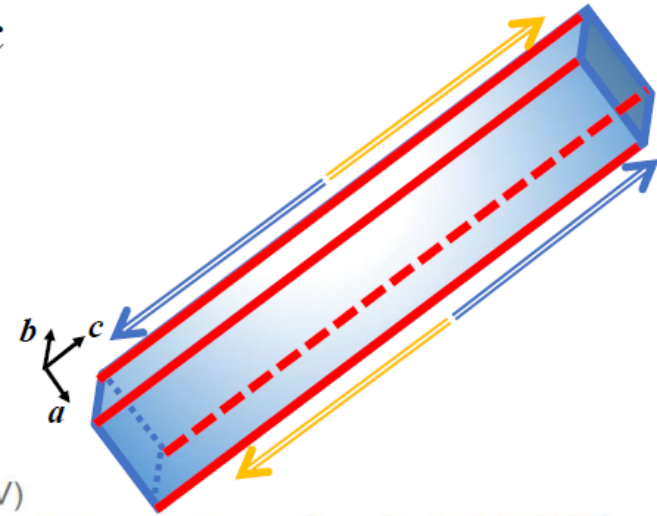
They all vanish

**So hinge state!**

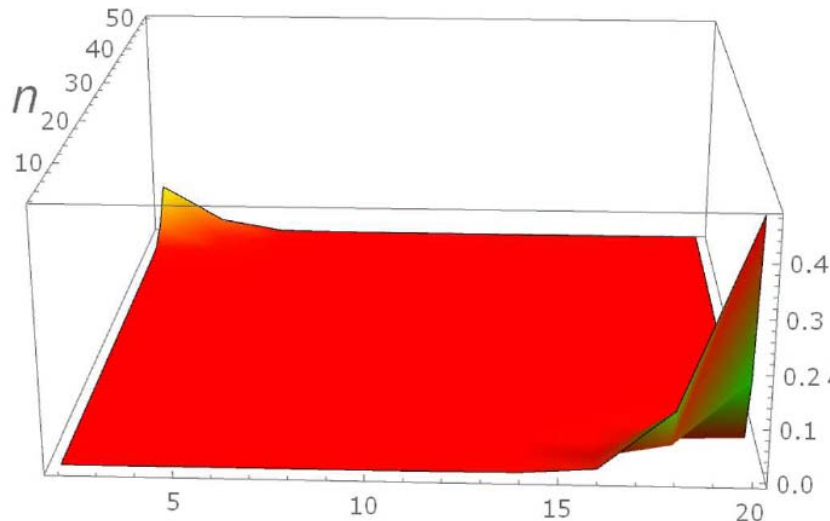
# Hinge state from TB calculation



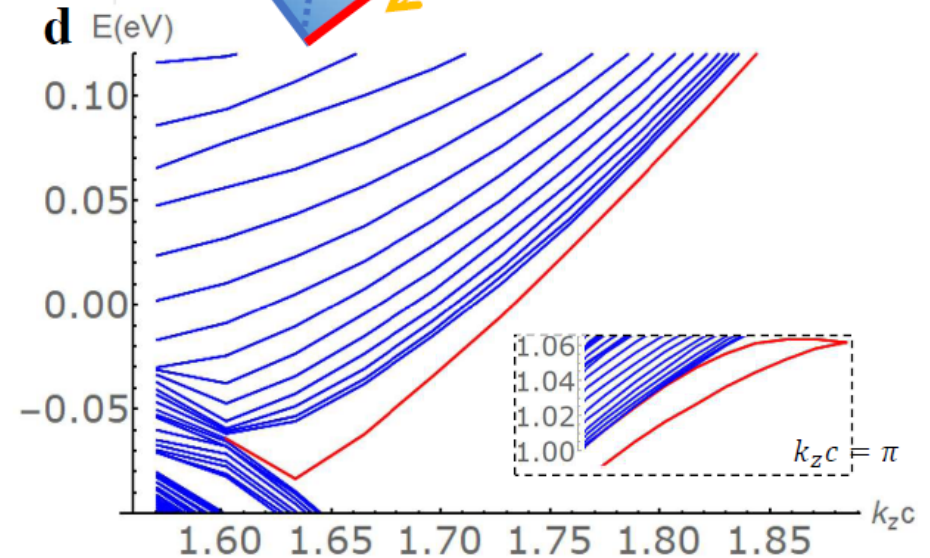
c



TB reproduces topo-feature and band dispersion



d



*Had been confirmed by  
Wang, Wieder, Li, Yan & Bernevig,  
arXiv:1806.11116 (2018)*

# BiBr: Dirac Surface States Coexisting with Hinge State

$SG12 (C2/m)$

TABLE III. For  $SG12$ , the HSPs are given by the labels  $\Gamma, A, \dots$  in order, and their coordinates can be referred to Ref. [2]. For the labeling of the irreps of  $\mathcal{G}(\mathbf{k}_i)$ , we use  $(j, m)$  where  $j$  means the  $j$ th irrep as listed in order by Ref. [2] and  $m$  denotes the dimension. The red color means that due to  $\mathcal{T}$ , the irrep must occur simultaneously with its  $\mathcal{T}$  pair which belongs to the same irrep.

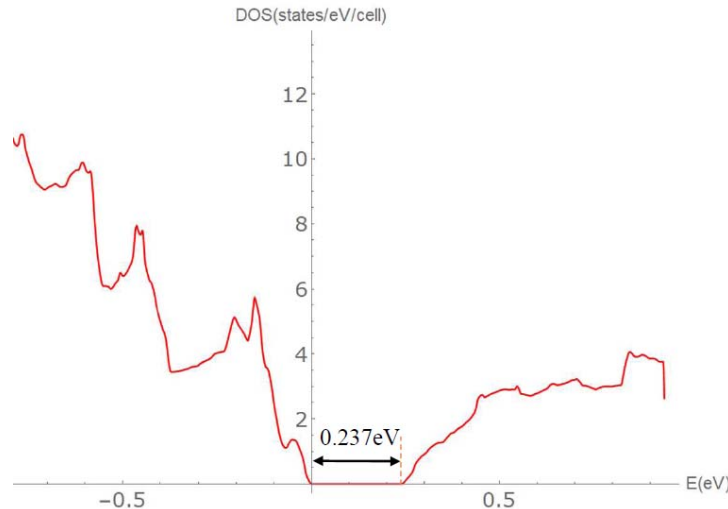
HSP	$\Gamma$				$A$				$Z$				$M$				$L$		$V$	
irrep	(1,1)	(2,1)	(3,1)	(4,1)	(1,1)	(2,1)	(3,1)	(4,1)	(1,1)	(2,1)	(3,1)	(4,1)	(1,1)	(2,1)	(3,1)	(4,1)	(1,1)	(2,1)	(1,1)	(2,1)
$n_{\mathbf{k}_i}^{\alpha_i}$	18	18	14	14	16	16	16	16	16	16	16	16	16	16	16	16	16	16	16	16

$$\mathbb{Z}_2 \times \mathbb{Z}_2 \times \mathbb{Z}_4$$

$$\mathbf{n} = (64, 18, 18, 14, 14, 16, 16, 16, 16, 16, 16, 16, 16, 16, 16, 16, 16, 16)$$

$$= 14\mathbf{a}_1 + 2\mathbf{a}_2 + 2\mathbf{a}_3 + 2\mathbf{a}_4 + \mathbf{a}_6 - \frac{1}{2}\mathbf{a}_7,$$

$$\text{SI of } (0, 0, 2)$$

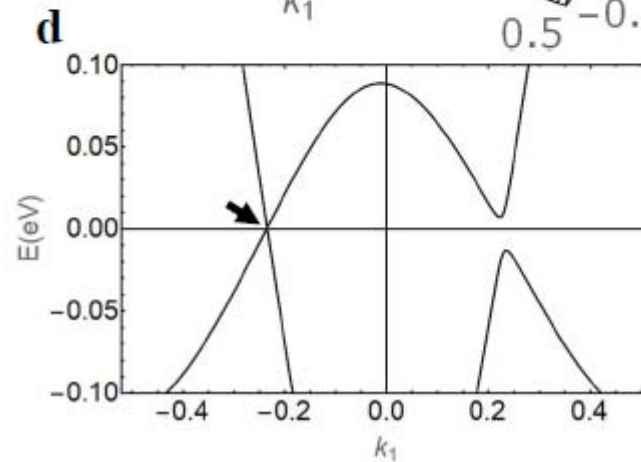
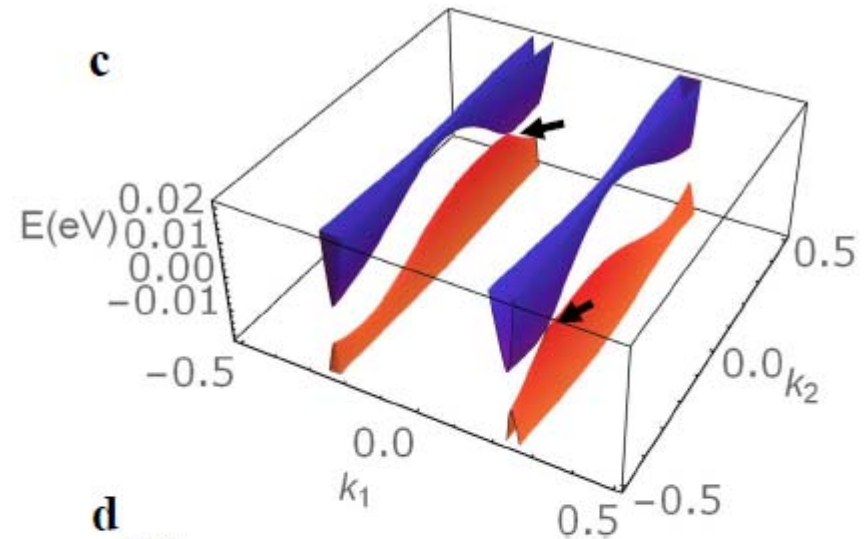
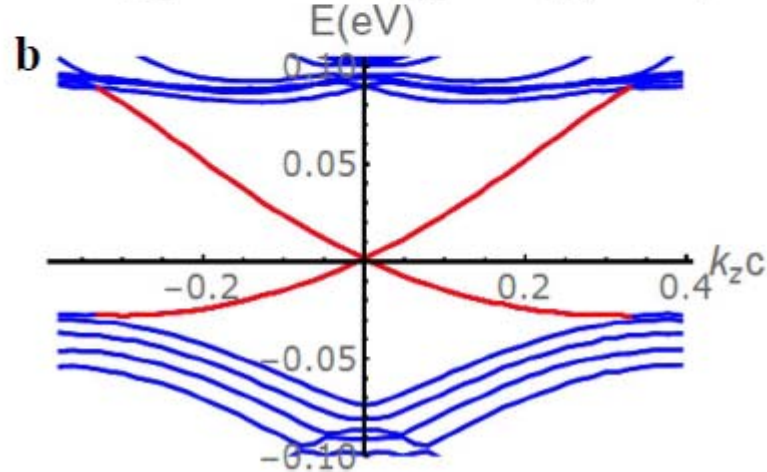
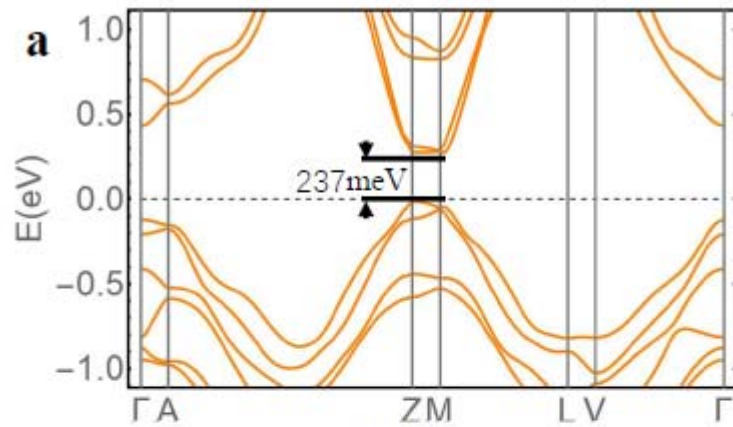


(c) The 7 AI basis vectors for  $SG12$ .

$SG12$	$\mathbf{a}_1$	$\mathbf{a}_2$	$\mathbf{a}_3$	$\mathbf{a}_4$	$\mathbf{a}_5$	$\mathbf{a}_6$	$\mathbf{a}_7$
$\nu$	4	4	-8	2	-4	8	-8
$\Gamma_1^1$	1	2	-4	1	-2	4	-4
$\Gamma_2^1$	1	2	-4	1	-2	4	-4
$\Gamma_3^1$	1	0	0	0	0	0	0
$\Gamma_4^1$	1	0	0	0	0	0	0
$A_1^1$	1	0	-1	0	0	2	-4
$A_2^1$	1	0	-1	0	0	2	-4
$A_3^1$	1	2	-3	1	-2	2	0
$A_4^1$	1	2	-3	1	-2	2	0
$Z_1^1$	1	0	-2	1	0	2	-4
$Z_2^1$	1	0	-2	1	0	2	-4
$Z_3^1$	1	2	-2	0	-2	2	0
$Z_4^1$	1	2	-2	0	-2	2	0
$M_1^1$	1	2	-3	0	-2	4	0
$M_2^1$	1	2	-3	0	-2	4	0
$M_3^1$	1	0	-1	1	0	0	-4
$M_4^1$	1	0	-1	1	0	0	-4
$L_1^1$	1	1	-1	1	-2	0	0
$L_2^1$	1	1	-3	0	0	4	-4
$V_1^1$	1	1	0	0	0	0	0
$V_2^1$	1	1	-4	1	-2	4	-4



# Topological states in BiBr



- Mirror Chern number = 0
- Hinge-State + Surface Dirac State
- TB  $\rightarrow$  topo-feature

Confirmed by:  
Hsu C H *et al.* 2D Mater., 2019, 6:031004

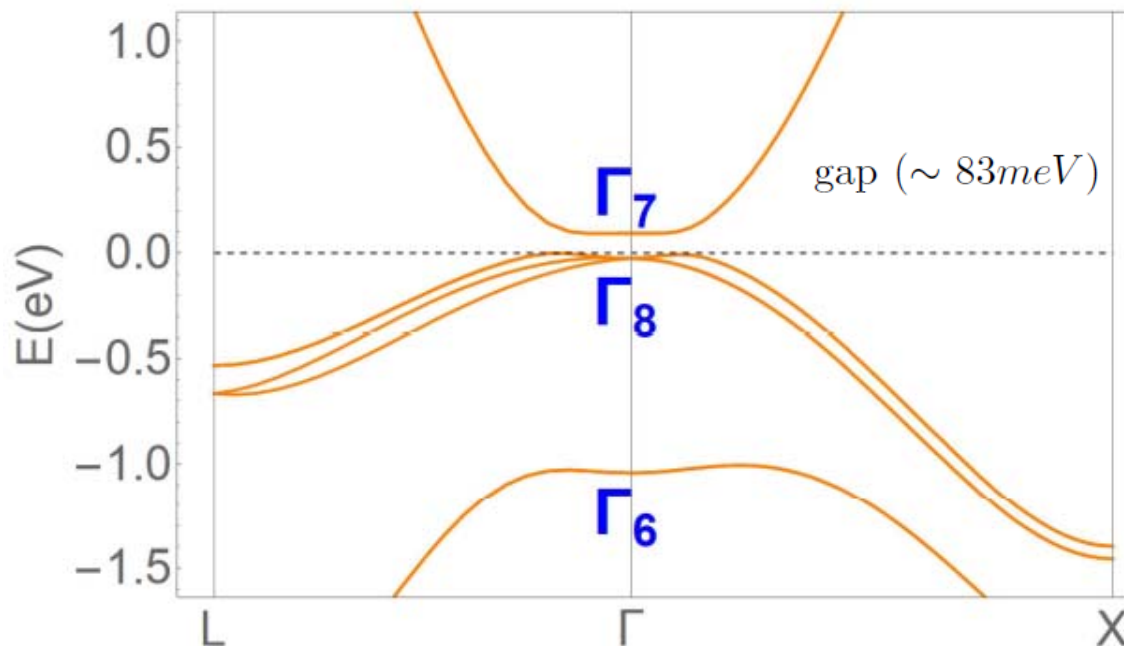


# Non-Centrosymmetric Strong TI in AgNaO

- Inversion  $\rightarrow$  Fu-Kane parity criterion
- Non-inversion  $\rightarrow$  calculate  $Z_2$  invariant

SG 216 ( $F\bar{4}3m$ ), which are non-centrosymmetric and contains  $S_4$

Expansion  $\rightarrow$  be insulators with the SI 1  $\mathbb{Z}_2$



$s(d)$  band is mainly above (below) the Fermi level.

band inversion near the Gamma point,

results in an  $S_4$  invariant

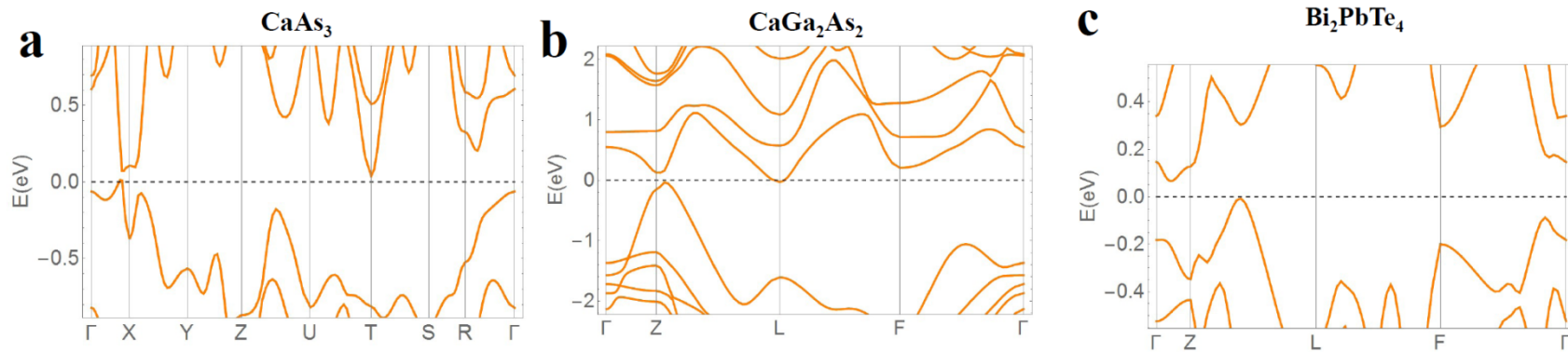
# Other $2 \in \mathbb{Z}_4$ Materials

- Cubic crystal  $\text{TiS}_2$  (SG227) is a glide-protected TCI with **hourglass** surface states
- Elemental phosphorus in the A7 structure (SG166), which occurs at about 9GPa, is predicted to be an inversion-protected TCI with 1D **hinge** states
- $\text{Ag}_2\text{F}_5$  (SG2) is a **weak TI** with additional inversion-protected band topology characterized by the invariant  $\kappa_1=2$

# Strong TI Found by 1,3 of $\mathbb{Z}_4$

TABLE V. Table of centrosymmetric STI candidates discovered by 1, 3 in  $\mathbb{Z}_4$

$SG$	Material	$X_{BS}$	SI	$(\nu_0; \nu_1, \nu_2, \nu_3)$	$\kappa_1$
2	CaAs <sub>3</sub> [17]	$\mathbb{Z}_2 \times \mathbb{Z}_2 \times \mathbb{Z}_2 \times \mathbb{Z}_4$	(0,0,1,1)	(1;1,0,0)	1
166	Bi <sub>2</sub> PbTe <sub>4</sub> [18]	$\mathbb{Z}_2 \times \mathbb{Z}_4$	(1,1)	(1;1,1,1)	3
166	CaGa <sub>2</sub> As <sub>2</sub> [19]	$\mathbb{Z}_2 \times \mathbb{Z}_4$	(1,1)	(1;1,1,1)	1

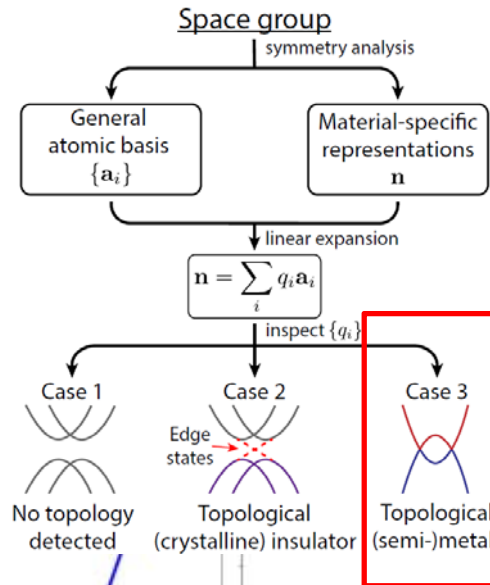


# Dirac Semimetal $\text{MgBi}_2\text{O}_6$

$SG136$

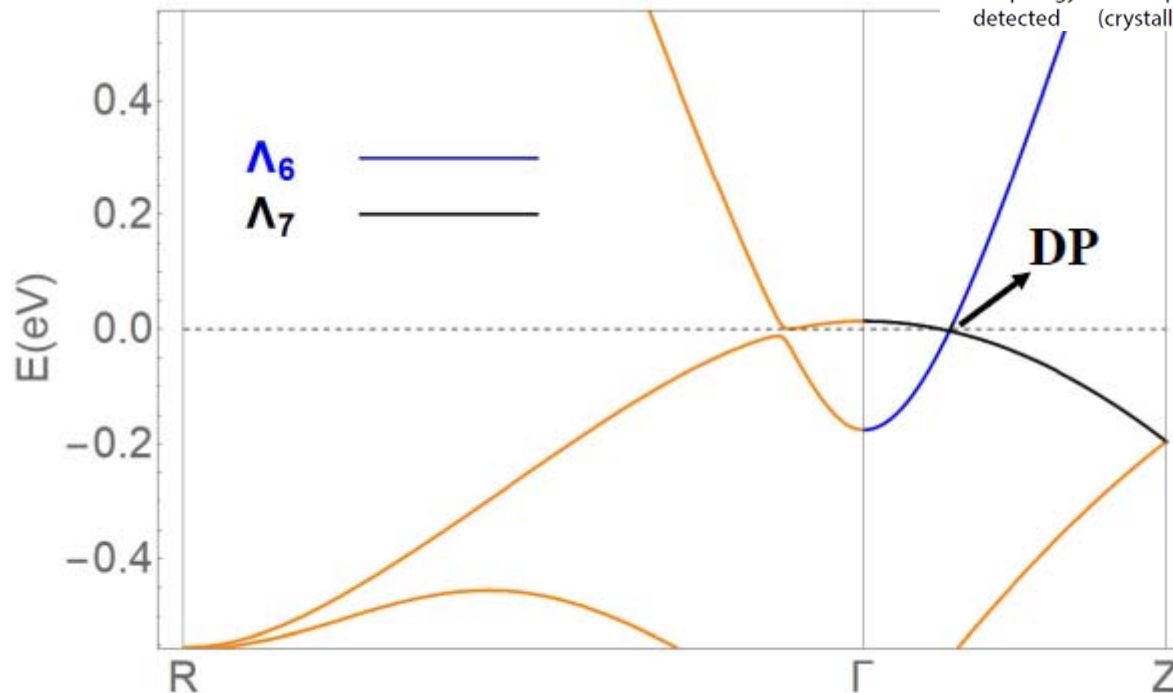
Expansion  $\rightarrow$  case 3

So topological Semimetal



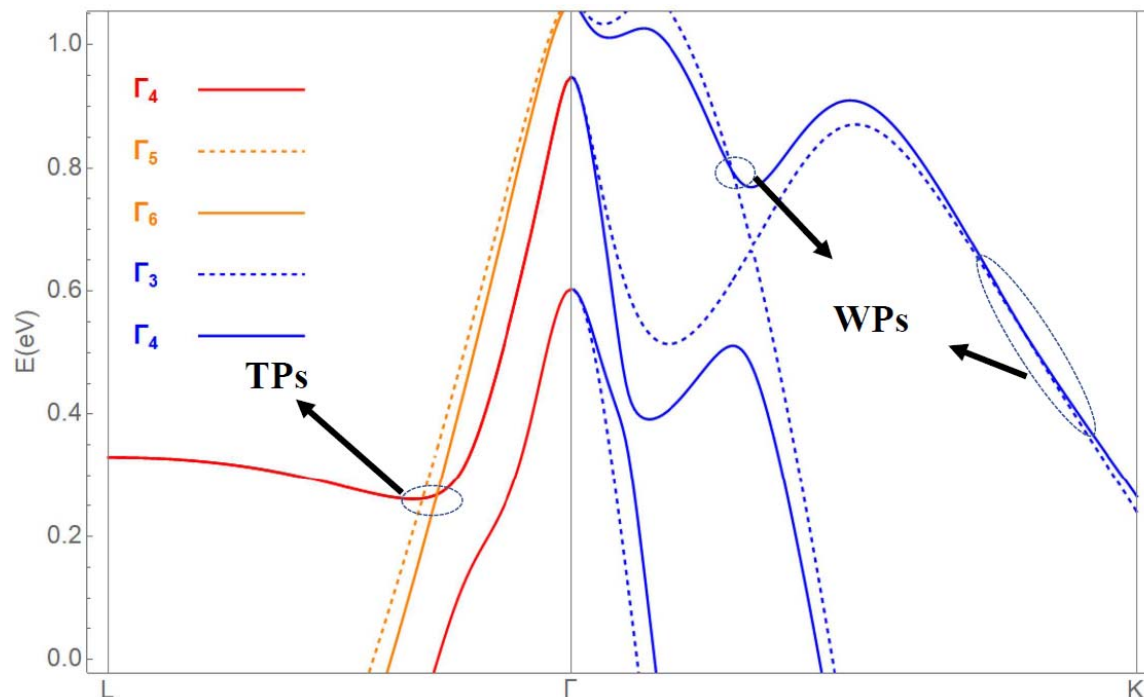
e) The 8 AI basis vectors for  $SG136$ .

$SG136$	$a_1$	$a_2$	$a_3$	$a_4$	$a_5$
$\nu$	8	8	8	4	-48
$\Gamma_1^2$	1	2	2	1	-12
$\Gamma_2^2$	1	0	2	1	-8
$\Gamma_3^2$	1	0	0	0	0
$\Gamma_4^2$	1	2	0	0	-4
$K_1^2$	1	1	0	1	-4
$K_2^2$	1	1	0	1	-4
$K_3^2$	1	1	2	0	-8
$K_4^2$	1	1	2	0	-8
$K_1^4$	2	2	2	1	-12
$K_1^4$	2	2	2	1	-12
$K_1^1$	1	1	1	0	-4
$K_2^1$	1	1	1	0	-4
$K_3^1$	1	1	1	0	-4
$K_4^1$	1	1	1	0	-4
$K_5^1$	1	1	1	1	-8
$K_6^1$	1	1	1	1	-8
$K_7^1$	1	1	1	1	-8
$K_8^1$	1	1	1	1	-8
$K_1^2$	2	2	2	1	-12
$K_2^2$	2	2	2	1	-12



# Three-Fold Degenerate Fermions

- AuLiMgSn (SG216)
- DFT  $\rightarrow n_{\mathbf{k}}^{\alpha}$  are all integers thus there are finite direct gaps in all the HSPs.
- However expansion on the SG216's AI basis vectors shows that they cannot constitute a BS at all, namely case 3 in the main text



# The Nodal-line Semimetal AgF<sub>2</sub>

AgF<sub>2</sub> (SG61)

$$\left( \uparrow \frac{11}{2}, \frac{19}{2}, \frac{13}{4} \right)$$

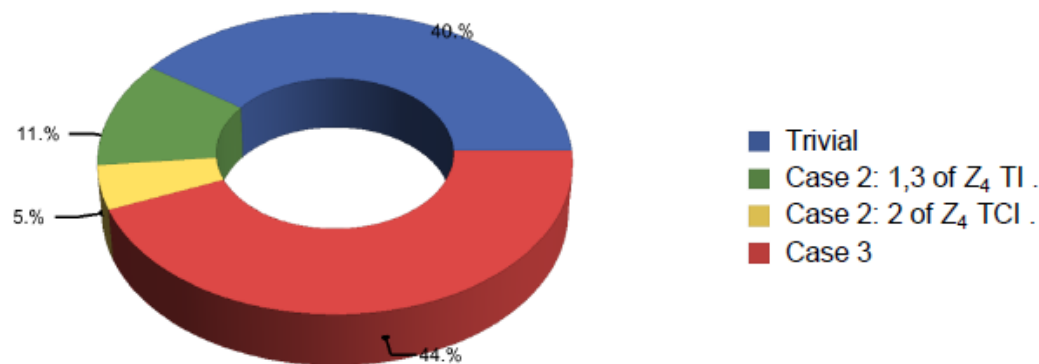
(d) The 3 AI basis vectors for SG61.

SG61	a <sub>1</sub>	a <sub>2</sub>	a <sub>3</sub>
$\nu$	16	8	-32
$\Gamma_1^2$	4	4	-12
$\Gamma_2^2$	4	0	-4
$Y_1^2$	4	2	-8
$Y_2^2$	4	2	-8
$X_1^2$	4	2	-8
$X_2^2$	4	2	-8
$Z_1^2$	4	2	-8
$Z_2^2$	4	2	-8
$U_1^2$	2	1	-4
$U_2^2$	2	1	-4
$T_1^2$	2	1	-4
$T_2^2$	2	1	-4
$S_1^2$	2	1	-4
$S_2^2$	2	1	-4
$R_1^1$	1	0	0
$R_2^1$	1	0	0
$R_3^1$	1	0	0
$R_4^1$	1	0	0
$R_5^1$	1	1	-4
$R_6^1$	1	1	-4
$R_7^1$	1	1	-4
$R_8^1$	1	1	-4

# Comprehensive database searches

- Around 180000 compounds at ICSD database
- Consider stoichiometric compound
- Do not consider  $3d/4f/5f$  and several  $4d/5d$  element
- ➔ **19143** compounds with SOC+time-reverse symmetry
  
- Highly efficient (main jobs is around 1 month within a cluster 28-node/one-node-32 CPU core)

GGA calculation  
Check by MBJ calculation





# List 258 good topological insulator

[ccmp.nju.edu.cn](http://ccmp.nju.edu.cn)

SG	$X_{BS}$	Topological insulators
2	$Z_2^3 \times Z_4$	AgO <sub>4</sub> S[3], <b>Ag<sub>3</sub>F<sub>5</sub></b> [4], As <sub>3</sub> Ca[5, 6], As <sub>3</sub> Sr[7, 8], Bi[9], Br <sub>5</sub> W[10], CaMo <sub>6</sub> S <sub>8</sub> [11], CaP <sub>3</sub> [8, 12]
11	$Z_2^3 \times Z_4$	BBeLi[13], GeHPd[14], HfS <sub>3</sub> [15], Mo <sub>2</sub> S <sub>3</sub> [16], NbPt <sub>3</sub> [17], Nb <sub>2</sub> Se <sub>3</sub> [18], Se <sub>3</sub> Ta <sub>2</sub> [18, 19]
12	$Z_2^3 \times Z_4$	Ag <sub>4</sub> K <sub>2</sub> Se <sub>3</sub> [20], Al <sub>2</sub> Ge <sub>4</sub> Sr <sub>3</sub> [21], Al <sub>5</sub> Mo <sub>3</sub> [22], As <sub>3</sub> Ba[8, 23], As <sub>3</sub> Sr[8, 24], Au <sub>2</sub> P <sub>3</sub> [25], BaSb <sub>3</sub> [26] Ba <sub>2</sub> Cd <sub>2</sub> Sb <sub>4</sub> [27], Ba <sub>2</sub> Hg <sub>3</sub> O <sub>14</sub> Pd <sub>7</sub> [28], Ba <sub>3</sub> Li <sub>4</sub> Sn <sub>3</sub> [29], <b>BaTi<sub>4</sub></b> [30], Ba <sub>11</sub> Bi <sub>14</sub> Cd <sub>8</sub> [31], Bi <sub>2</sub> Pb <sub>3</sub> Se <sub>6</sub> [32], Bi <sub>2</sub> Pd[33] Bi <sub>4</sub> Pb <sub>7</sub> Se <sub>13</sub> [34], Br <sub>4</sub> Cs <sub>2</sub> I <sub>3</sub> Pd[35], <b>Hg<sub>2</sub>Sn<sub>2</sub>Sr</b> [36], Nb <sub>2</sub> PdS <sub>5</sub> [37], Nb <sub>2</sub> PdSe <sub>6</sub> [38], <b>Nb<sub>2</sub>PdSe<sub>6</sub></b> [38], Pa <sub>3</sub> Sr[39] , PdSe <sub>6</sub> Ta <sub>2</sub> [38], PtSe <sub>7</sub> Ta <sub>2</sub> [40]
14	$Z_2 \times Z_4$	Ag <sub>2</sub> Te[41, 42]
51	$Z_2^3 \times Z_4$	AlPt <sub>2</sub> [43], AuTe <sub>3</sub> [44], Au <sub>3</sub> Rb <sub>2</sub> Tl[45]
55	$Z_2 \times Z_4$	<b>Al<sub>3</sub>Bi<sub>6</sub>Ca<sub>6</sub></b> [46], <b>Bi<sub>6</sub>In<sub>2</sub>Sr<sub>5</sub></b> [47]
57	$Z_2^3 \times Z_4$	STa <sub>2</sub> [48]
58	$Z_4$	Bi <sub>3</sub> Hf[49], S <sub>2</sub> Ti[50]
59	$Z_2 \times Z_4$	Ag <sub>3</sub> Sb[51], AuCs <sub>3</sub> Pb <sub>4</sub> [52], AuPb <sub>4</sub> Rb <sub>3</sub> [52], AuRb <sub>3</sub> Sn <sub>4</sub> [52]
60	$Z_4$	Au <sub>2</sub> Pb[53, 54], O <sub>2</sub> Pb[55]
61	$Z_4$	AuSn <sub>2</sub> [56], Bi <sub>3</sub> Pt[57]
62	$Z_4$	AsCdNa[58], As <sub>2</sub> Hf[59], As <sub>2</sub> Zr[60, 61], Ba <sub>2</sub> Pb[62], Ba <sub>2</sub> Si[63, 64], Ca <sub>3</sub> GeO[65], CdGeSr[66], CdNaSb[67, 68], FSeY[69] GaPtSc[70], GaPtY[70], GeHPt[71, 72], GeMoZr[73], GePdZr[74, 75], GePtZr[75, 76], GeZr[77], Ge <sub>2</sub> InLiSr <sub>2</sub> [78], Ge <sub>2</sub> Mo[79] Ge <sub>2</sub> W[80], HfNbP[81], HfP <sub>3</sub> [82], HfSi[83], NNa <sub>3</sub> [84], N <sub>3</sub> Ta <sub>3</sub> [85], NbPZr[86], PPTc[75, 87], P <sub>2</sub> Ti[88] P <sub>2</sub> Zr[89], PbSr <sub>2</sub> [90, 91], PdSiTi[75, 92], PdSiZr[93], PtSiTi[75, 94], PtSiZr[75, 94], Sb <sub>2</sub> SrZn[95, 96], SiSr[97], SIT[98], SiZr[98]
63	$Z_2 \times Z_4$	Ag <sub>3</sub> Ta[99], AlBMo[100], AlBW[100], Al <sub>2</sub> BaSi <sub>2</sub> [101], AuClTe <sub>2</sub> [102], B <sub>7</sub> WY <sub>3</sub> [103], BaIn[104], CaSi [105] Ga[106], GeNa <sub>2</sub> Zn[107], HfPd[108], HfTe <sub>6</sub> [109, 110], PdZr[111]
64	$Z_2 \times Z_4$	As[112], C <sub>2</sub> B <sub>2</sub> Mg[113], LiTi[114]
65	$Z_2^3 \times Z_4$	<b>Ag<sub>3</sub>Te<sub>2</sub>Tl</b> [115], Ba <sub>3</sub> Ge <sub>4</sub> [116]
69	$Z_2^3 \times Z_4$	Be <sub>2</sub> Zn[117]
71	$Z_2 \times Z_4$	<b>AsTeTi</b> [118], Br <sub>2</sub> Ca <sub>3</sub> Si[119]
72	$Z_2 \times Z_4$	Br <sub>2</sub> Hg <sub>5</sub> O <sub>4</sub> [120]
87	$Z_2 \times Z_8$	As <sub>4</sub> Mo <sub>6</sub> [121], Hf <sub>5</sub> Te <sub>4</sub> [122], Se <sub>4</sub> Ti <sub>5</sub> [123], Te <sub>4</sub> Zr <sub>5</sub> [124]
114	$Z_2$	Pd <sub>4</sub> S[125]
121	$Z_2$	Ag <sub>2</sub> S <sub>4</sub> SnZn[126]
122	$Z_2$	As <sub>2</sub> CdGe[127], As <sub>2</sub> CdSn[128], As <sub>2</sub> SnZn[129], CdSb <sub>2</sub> Sn[130]
123	$Z_2 \times Z_4 \times Z_8$	BaGe <sub>2</sub> Mg <sub>4</sub> [131], <b>BiLi</b> [132], BiNa[133], ClNa <sub>2</sub> [134], ClNa <sub>3</sub> [134]
127	$Z_4 \times Z_8$	B <sub>4</sub> Y[135], CsI <sub>3</sub> Sn[136, 137], Pt <sub>3</sub> Si[138]
129	$Z_2 \times Z_4$	HfSb <sub>2</sub> [139]
136	$Z_4$	Ag <sub>5</sub> CsSe <sub>3</sub> [140], Ag <sub>5</sub> CsTe <sub>3</sub> [141]
139	$Z_2 \times Z_8$	Ag <sub>2</sub> Zr[142], Au <sub>2</sub> Hf[143], Au <sub>2</sub> Zr[144], BaCd <sub>2</sub> Ge <sub>2</sub> [145], BaGe <sub>2</sub> Mg <sub>2</sub> [146], Be <sub>12</sub> W [147] Bi <sub>10</sub> Ca <sub>11</sub> [148], CaGe <sub>2</sub> Zn <sub>2</sub> [146], Cd <sub>2</sub> Ge <sub>2</sub> Sr[149], Ge <sub>2</sub> SrZn <sub>2</sub> [150], Hf <sub>2</sub> Pd[151], InPd <sub>3</sub> [152], PdZr <sub>2</sub> [153], Pd <sub>2</sub> Ti[154]
140	$Z_2 \times Z_8$	Bi <sub>3</sub> In <sub>3</sub> [155], Hf <sub>2</sub> Si[156], In <sub>3</sub> Sb <sub>2</sub> [157], Pb <sub>2</sub> Pt[158], Sc[159], SiZr <sub>2</sub> [160], <b>Sb<sub>2</sub>Tl</b> [161], Si <sub>3</sub> Sr <sub>5</sub> [162], Sn <sub>3</sub> Sr <sub>5</sub> [163]
164	$Z_2 \times Z_4$	Ag <sub>2</sub> O[164], As <sub>2</sub> Ge <sub>2</sub> Te <sub>3</sub> [165], As <sub>4</sub> GeTe <sub>7</sub> [166], BaSn <sub>2</sub> [167, 168], Be[169], Bi <sub>2</sub> Li <sub>3</sub> Y[170], Bi <sub>2</sub> Mg <sub>3</sub> [171, 172] Bi <sub>2</sub> Pb <sub>2</sub> Se <sub>6</sub> [173, 174], Bi <sub>4</sub> SnTe <sub>7</sub> [175], CNb <sub>2</sub> [176], CTa <sub>2</sub> [177], CeK <sub>2</sub> N <sub>6</sub> PtS <sub>6</sub> [178], CeN <sub>6</sub> PrRb <sub>2</sub> S <sub>6</sub> [178] CaGe <sub>2</sub> [179, 180], CaSi <sub>2</sub> [180, 181], GeSb <sub>4</sub> Te <sub>7</sub> [182], Ge <sub>2</sub> Sr[180, 183], PdTe <sub>2</sub> [184, 185], PtTe <sub>2</sub> [186, 187], Sb <sub>2</sub> Te <sub>3</sub> [188], SiTe <sub>2</sub> [189]
166	$Z_2 \times Z_4$	CsGa <sub>7</sub> [190], Cl <sub>3</sub> Na <sub>2</sub> Tl <sub>3</sub> [191], <b>Ga<sub>7</sub>Rb</b> [192], PtTe[193], Pt <sub>2</sub> Te <sub>3</sub> [193], STi[194], SeTi[195], AgGeLi <sub>2</sub> [196] Al <sub>10</sub> Ba <sub>7</sub> [197], As[198], As <sub>2</sub> CaGa <sub>2</sub> [199] As <sub>2</sub> Ge <sub>4</sub> Te <sub>7</sub> [200], As <sub>2</sub> Sn <sub>2</sub> Sr[201, 202], Bi <sub>2</sub> GeTe <sub>4</sub> [203, 204], Bi <sub>2</sub> PbTe <sub>4</sub> [205], Bi <sub>2</sub> SeTe <sub>2</sub> [206, 207], Bi <sub>2</sub> Se <sub>3</sub> [208, 209], Bi <sub>2</sub> Te <sub>3</sub> [209, 210] Bi <sub>6</sub> PbTe <sub>10</sub> [210], Bi <sub>8</sub> Te <sub>9</sub> [211], <b>CS<sub>2</sub>Ta<sub>2</sub></b> [212], CY <sub>2</sub> [213], C <sub>5</sub> Ti <sub>8</sub> [214] CaGe <sub>2</sub> [215], CaSi <sub>2</sub> [216, 217], GaGeTe[218] GaP[219], Ga <sub>2</sub> Te <sub>3</sub> [220], In <sub>2</sub> Te <sub>3</sub> [220], Li <sub>3</sub> Pb <sub>3</sub> [221] N <sub>2</sub> W[222], Sb[223], SbTe <sub>2</sub> Tl[224, 225], Sb <sub>2</sub> SeTe <sub>2</sub> [226, 227] Sb <sub>2</sub> SnTe <sub>4</sub> [228, 229], Sb <sub>2</sub> Te <sub>3</sub> [209, 230], Sb <sub>5</sub> Te <sub>3</sub> [231], Sb <sub>48</sub> Te <sub>9</sub> [231]
191	$Z_6 \times Z_{12}$	B <sub>2</sub> Sr[232], BaSi <sub>2</sub> [233], CaSi <sub>2</sub> [234], Li <sub>2</sub> Pd[235], Li <sub>2</sub> Pt[236], Ti[237]
193	$Z_{12}$	Pb <sub>3</sub> TeZr <sub>5</sub> [238], Sn <sub>3</sub> Zr <sub>5</sub> [239]
194	$Z_{12}$	AgSbSr[240], AlN <sub>3</sub> Ti <sub>4</sub> [241], Al <sub>2</sub> Hf[242], AuBaBi[243], C <sub>2</sub> AlIa <sub>3</sub> [244], Ca <sub>2</sub> IN[245] MgPo[246], PbPt[247], SiSr <sub>2</sub> [248], BeHfSi[249]
216	$Z_2$	AgKO[250], AgNaO[250], AgORb[250], HgS[251], InN[252], NTI [253]
221	$Z_4 \times Z_8$	AlSc[254], AlY[255], BeCa[256], BeSr[256], BeTi[257] Br <sub>3</sub> CsGe[258], Br <sub>3</sub> CsPb[259], CsI <sub>3</sub> Sn[260] MgPd[261], PbSe[262], SnTe[263]
223	$Z_4$	BiNb <sub>3</sub> [264], Nb <sub>3</sub> Sb[265], SbIa <sub>3</sub> [266, 267]
225	$Z_8$	BiY[268], CPd[269], CZr[270], Li <sub>2</sub> MgSi[271], N <sub>2</sub> Pd[272]
227	$Z_4$	CTl <sub>2</sub> [273], Se[274]

TABLE I. The list of all the TIs with relatively clean Fermi surfaces discovered by GGA calculations: when the MBJ calculation doesn't change the band topology predicted by GGA calculations, we print the material in blue, while the red color means that the MBJ calculation results in a topological crystalline phase.

$SG$	$X_{BS}$	Topological crystalline insulators
2	$Z_3^2 \times Z_4$	CsHg[275],HgK[276]
11	$Z_2^2 \times Z_4$	BaSb <sub>2</sub> [277],MoTe <sub>2</sub> [278],Sb <sub>2</sub> Sr[279],Se <sub>3</sub> Ta[280, 281]
12	$Z_2^3 \times Z_4$	Al <sub>4</sub> Cl <sub>5</sub> Zr <sub>12</sub> [282],Al <sub>4</sub> Na <sub>4</sub> P <sub>12</sub> Sr <sub>8</sub> [283],As <sub>2</sub> Nb[284, 285],As <sub>2</sub> Ta[285, 286],As <sub>3</sub> Mo <sub>2</sub> [287],As <sub>3</sub> W <sub>2</sub> [288] As <sub>4</sub> Ba <sub>3</sub> Zn <sub>2</sub> [289],Ba <sub>3</sub> Cd <sub>2</sub> Sb <sub>4</sub> [290],BiBr[291],BiHf <sub>2</sub> [292],Bi <sub>3</sub> ITe[293, 294],CdK <sub>6</sub> Pb <sub>8</sub> [295],ClZr [296] Cl <sub>8</sub> NSc <sub>5</sub> [297],NbP <sub>2</sub> [298],NbSb <sub>2</sub> [299],Sb <sub>2</sub> Ta[300, 301],Se <sub>4</sub> Ti <sub>3</sub> [302],Ta <sub>2</sub> Te <sub>3</sub> [303]
51	$Z_2^2 \times Z_4$	GaPt <sub>2</sub> [304],CdMg[305]
55	$Z_2 \times Z_4$	Ca <sub>5</sub> Ga <sub>2</sub> Sb <sub>6</sub> [306]
58	$Z_4$	Bi <sub>3</sub> RbS <sub>5</sub> [307]
59	$Z_2 \times Z_4$	BrNTi[308]
62	$Z_4$	HgSr <sub>3</sub> [309],PdSi[310],PdSn[311]
63	$Z_2 \times Z_4$	BaGe[312],BaSi[313], BaSn[314],Ba <sub>3</sub> Pb <sub>5</sub> [315],Ba <sub>5</sub> Cd <sub>2</sub> FSb <sub>5</sub> [316], CaGe[317],CaSn [318] Cd <sub>2</sub> FSb <sub>5</sub> Sr <sub>5</sub> [319],PbSr[320],SnSr[321]
64	$Z_2 \times Z_4$	Ca <sub>5</sub> Ga <sub>2</sub> N <sub>4</sub> [322],Li[323]
65	$Z_2^2 \times Z_4$	Au <sub>7</sub> Rb <sub>3</sub> [324]
69	$Z_2^2 \times Z_4$	Ge <sub>6</sub> Li <sub>2</sub> Sr <sub>4</sub> [325]
71	$Z_2 \times Z_4$	Ba <sub>3</sub> Bi <sub>4</sub> Li <sub>4</sub> [326],Ba <sub>3</sub> Li <sub>4</sub> Sb <sub>4</sub> [326]
87	$Z_2 \times Z_8$	Au <sub>4</sub> Ti[327]
88	$Z_4$	O <sub>4</sub> PbPd <sub>2</sub> [328]
123	$Z_2 \times Z_4 \times Z_8$	As <sub>3</sub> CsZn <sub>4</sub> [329],As <sub>3</sub> RbZn <sub>4</sub> [329],CPd <sub>3</sub> Sn[330], Pd <sub>3</sub> Sn[331]
127	$Z_4 \times Z_8$	B <sub>4</sub> Ca[332]
129	$Z_2 \times Z_4$	AsGeNb[333],GeHfS[334, 335],GeHfSe[334, 335],GeHfTe[334, 335],GeNbSb[336],GeSZr [337] GeSeZr[334],HfSSi[334, 335],HfSeSi[334, 335],OSiZr[334],SSiZr[338],SeSiZr[338],SiTeZr[338],SnTeZr[338]
137	$Z_4$	Ba <sub>2</sub> LiN[339]
139	$Z_2 \times Z_8$	AsBa <sub>2</sub> [340],AsCa <sub>2</sub> [341] Ba <sub>2</sub> Bi[342],Ba <sub>2</sub> Sb[343],Ba <sub>11</sub> Bi <sub>10</sub> [344],BiSr <sub>2</sub> [343],Bi <sub>2</sub> F <sub>2</sub> OSr <sub>2</sub> Ti <sub>2</sub> [345],Br <sub>2</sub> Ca <sub>3</sub> Si [346] CaIn[347],F <sub>2</sub> OSb <sub>2</sub> Sr <sub>2</sub> Ti <sub>2</sub> [348],Hf <sub>2</sub> Hg[349],Pd <sub>2</sub> Si <sub>2</sub> Sr[350],SbSr <sub>2</sub> [351]
140	$Z_2 \times Z_8$	Bi[352],GePt <sub>3</sub> [353],Mg <sub>2</sub> Pt[354]
164	$Z_2 \times Z_4$	BaSi <sub>2</sub> [355],BiTe[356],Bi <sub>2</sub> Pb <sub>2</sub> Te <sub>5</sub> [357, 358], Bi <sub>2</sub> Se <sub>2</sub> [359],Bi <sub>3</sub> STe <sub>2</sub> [360],HgPt <sub>2</sub> Se <sub>3</sub> [361],Te <sub>2</sub> Zr[362]
166	$Z_2 \times Z_4$	B <sub>2</sub> Mo[363],Ba <sub>3</sub> O <sub>9</sub> Sc <sub>4</sub> [364],Pt <sub>3</sub> Te <sub>4</sub> [365],TeZr <sub>3</sub> [366],AsNaTe <sub>2</sub> Zr <sub>2</sub> [367],As <sub>3</sub> Cd <sub>4</sub> K[368] As <sub>3</sub> Cd <sub>4</sub> Na[368],As <sub>3</sub> Cd <sub>4</sub> Rb[368],Bi[369],Bi <sub>4</sub> Se <sub>3</sub> [370, 371],Bi <sub>8</sub> Se <sub>9</sub> [372],BrZr[373],CNb <sub>2</sub> S <sub>2</sub> [374],C <sub>3</sub> Al <sub>6</sub> N <sub>2</sub> [375] ClZr[376],Hg[377],P[378],Pd <sub>3</sub> S <sub>2</sub> Tl <sub>2</sub> [379]
187	$Z_3 \times Z_3$	Br <sub>2</sub> Ca <sub>3</sub> Si[380],CHI[381],InNbS <sub>2</sub> [382, 383],InNbSe <sub>2</sub> [383, 384],NY[385]
189	$Z_3 \times Z_3$	AgAsCa[386, 387],AgCaP[387, 388] GeLiY[389]
191	$Z_6 \times Z_{12}$	B <sub>2</sub> Ca[390],B <sub>2</sub> Mg[391],B <sub>2</sub> Ti[392],B <sub>2</sub> Zr[393], Be <sub>12</sub> Ti[394],Ga <sub>2</sub> Sr[395],Pd <sub>5</sub> Sr[396]
193	$Z_{12}$	Pb <sub>3</sub> SZr <sub>5</sub> [397]
194	$Z_{12}$	AlLi[398],AsHgK[399, 400], B <sub>2</sub> W[401], Be[402],C[403],CdNa <sub>2</sub> Sn[404], HgKSb[400, 405],SnTi <sub>2</sub> [406]
221	$Z_4 \times Z_8$	CPbPd <sub>3</sub> [407],CPt <sub>3</sub> Sn[407],CaPd[408],Ca <sub>3</sub> GeO[409, 410],Ca <sub>3</sub> OPb[410, 411],HfPd <sub>3</sub> [412],HgPt <sub>3</sub> [413],NTi <sub>3</sub> Tl[414] NY[415],PbPt <sub>3</sub> [416],Pt <sub>3</sub> Sn[417],Pt <sub>3</sub> Zn[418]
225	$Z_8$	AsSc[419],Au <sub>2</sub> InSc[420],Au <sub>2</sub> InY[421],Ba[422],Bi <sub>3</sub> [423],BiSc[424] CPt[425],Ca[426],PbPo[427, 428],SSn[429],SbSc[430],SeSn[431], SnTe[432, 433]
227	$Z_4$	Mo <sub>2</sub> O <sub>7</sub> Y <sub>2</sub> [434],S <sub>2</sub> Ti[435]

List 165 good topological Crystalline insulator

<i>SG</i>	Topological (semi-)metals
11	Br <sub>9</sub> TeW[436], CBrHgNS[437], Li <sub>7</sub> Sn <sub>3</sub> [438], Mo <sub>2</sub> S <sub>2</sub> Sb[439]
51	AuCd[440], AuTi[441]
52	Ag <sub>2</sub> BiO <sub>3</sub> [442], Bi <sub>3</sub> Sr <sub>2</sub> [443]
55	Al <sub>3</sub> Pd <sub>5</sub> [444], Al <sub>3</sub> Pt <sub>5</sub> [445], BCl <sub>6</sub> Sc <sub>4</sub> [446], Bi <sub>9</sub> Ca <sub>9</sub> Cd <sub>4</sub> [447], Bi <sub>9</sub> Ca <sub>9</sub> Zn <sub>4</sub> [448], Bi <sub>9</sub> Cd <sub>4</sub> Sr <sub>9</sub> [447], In <sub>5</sub> S <sub>13</sub> Y <sub>4</sub> [449]
57	AlCaPd[450], BiK <sub>2</sub> Sn[451]
58	C <sub>60</sub> K[452]
59	Ag <sub>3</sub> Sn[453]
60	F <sub>4</sub> NaTi[454]
61	AgF <sub>2</sub> [455]
62	AgAuP <sub>7</sub> [456], AgF <sub>3</sub> K[457], AlPt <sub>2</sub> [458] Bi <sub>3</sub> Ca <sub>5</sub> [459], Bi <sub>3</sub> Sr <sub>5</sub> [460], Ca <sub>5</sub> Sb <sub>3</sub> [461] GeNb <sub>3</sub> Te <sub>6</sub> [462], GePd <sub>2</sub> Y[463], N <sub>3</sub> Nb <sub>2</sub> [464], Nb <sub>3</sub> Si[465] PdSbZr[466], SiTa <sub>3</sub> Te <sub>6</sub> [466]
63	AgCa[467], AuCa[468], BiZr[469], Ga <sub>3</sub> PdSr[470], Ga <sub>5</sub> Zr <sub>3</sub> [471], GeSc[472], GeY[473], HfSb[474], K <sub>3</sub> O <sub>4</sub> Pd <sub>2</sub> [475], K <sub>3</sub> O <sub>4</sub> Pt <sub>2</sub> [476], K <sub>4</sub> P <sub>3</sub> [477, 478], N <sub>5</sub> NaTa <sub>3</sub> [479], PdY[480], SiY[481], Sr <sub>3</sub> Tl <sub>5</sub> [482]
64	AgCs <sub>2</sub> F <sub>4</sub> [483], Au <sub>10</sub> Ca <sub>4</sub> In <sub>3</sub> [484], Bi[485, 486]
87	Ba <sub>9</sub> In <sub>4</sub> [487], In <sub>4</sub> Pd <sub>17</sub> Se <sub>4</sub> [488], Pt <sub>11</sub> Zr <sub>9</sub> [489], Pt <sub>12</sub> Si <sub>5</sub> [490]
88	Al <sub>21</sub> Pt <sub>8</sub> [491], CsFO <sub>3</sub> S[492], Ge <sub>8</sub> Pd <sub>21</sub> [493]
123	AgPPt <sub>5</sub> [494], AlPPt <sub>5</sub> [494], AsInPd <sub>5</sub> [495], AsPd <sub>5</sub> Tl[496] AsPt <sub>5</sub> Tl[496], As <sub>2</sub> BaPd <sub>2</sub> [497], BaP <sub>2</sub> Pd <sub>2</sub> [498], CaPb[499] CdPd[500], CdPd <sub>5</sub> Se[501], CdPt[502], Cd <sub>3</sub> Zr[503] FKNb <sub>4</sub> O <sub>5</sub> [504], HgPd[505], HgPd <sub>5</sub> Se[506], HgPt[507], Hg <sub>2</sub> Pt[507] InPPd <sub>5</sub> [508], InPPt <sub>5</sub> [508], PPd <sub>5</sub> Tl[508], PPT <sub>5</sub> Tl[508] PdTi[509], Pd <sub>5</sub> SeZn[510], SiSr[511], Sr[512]
127	AlSc <sub>2</sub> Si <sub>2</sub> [513], Au <sub>2</sub> Ca <sub>2</sub> Pb[514], Au <sub>2</sub> InY <sub>2</sub> [515], B <sub>2</sub> Ta <sub>3</sub> [516] B <sub>4</sub> W[517], C <sub>2</sub> B <sub>2</sub> Y[518], Ga <sub>2</sub> MgSc <sub>2</sub> [519], Ca <sub>3</sub> Hg <sub>2</sub> [520] Ga <sub>2</sub> Nb <sub>3</sub> [521], Ga <sub>2</sub> Ta <sub>3</sub> [522], Ge <sub>2</sub> Hf <sub>3</sub> [523], Hg <sub>2</sub> Sr <sub>3</sub> [524, 525] InPd <sub>2</sub> Y <sub>2</sub> [526], In <sub>2</sub> Ti <sub>2</sub> [527], LiSi <sub>2</sub> Y <sub>2</sub> [528], PbPd <sub>2</sub> Y <sub>2</sub> [529]
129	AgMgSb[530], AsNbSi[531], AsSiTa[532], BaMgSi[533], BiKMg[534], GeTeZr[535], MoNTa[536]
130	AlMg <sub>4</sub> Si <sub>3</sub> [537], Se <sub>3</sub> Tl <sub>5</sub> [538]
131	OPd[539]
136	AlNb <sub>2</sub> [540], AlTa <sub>2</sub> [541], Bi <sub>2</sub> MgO <sub>6</sub> [542], O <sub>2</sub> Pb[543, 544], O <sub>2</sub> Pd[545], O <sub>2</sub> Pt[546]
137	As <sub>2</sub> Cd <sub>3</sub> [547, 548], Bi <sub>2</sub> Se <sub>3</sub> [549]
139	Ag <sub>2</sub> CaGe <sub>2</sub> [550], Ag <sub>2</sub> Ge <sub>2</sub> Sr[550], Ag <sub>2</sub> Si <sub>2</sub> Sr[551], Ag <sub>2</sub> Sn <sub>2</sub> Sr[552], Al <sub>2</sub> BaSi <sub>2</sub> [553], Al <sub>2</sub> Pb <sub>2</sub> Sr[554], Al <sub>3</sub> Nb[555] Al <sub>3</sub> Ta[556], Al <sub>3</sub> Zr[557], Al <sub>4</sub> Ba[558], Au <sub>4</sub> CaCd <sub>2</sub> [559], BaBi <sub>4</sub> Cl <sub>2</sub> O <sub>6</sub> [560], BaIn <sub>4</sub> [561], Be <sub>12</sub> Pd[562], Be <sub>12</sub> Pt[562] CaGa <sub>4</sub> [563], Ca <sub>11</sub> Sb <sub>10</sub> [564], Ga <sub>3</sub> Nb[565, 566], Ga <sub>3</sub> Ta[567], In <sub>4</sub> Sr[568], Pt <sub>3</sub> Sb[569], S <sub>2</sub> Ti[570], Sb <sub>10</sub> Sr <sub>11</sub> [571] Si <sub>2</sub> W[572], ZnZr <sub>2</sub> [573]
140	AgCsF <sub>3</sub> [574], AgF <sub>3</sub> Rb[574], AgIn <sub>2</sub> [575], AlZr <sub>2</sub> [576], AuSe <sub>3</sub> Tl <sub>4</sub> [577], AuTl <sub>2</sub> [578], Au <sub>3</sub> Cd <sub>5</sub> [579] Au <sub>4</sub> In <sub>2</sub> K[580], Au <sub>4</sub> In <sub>2</sub> Rb[580] BGe <sub>2</sub> Nb <sub>5</sub> [581], BMo <sub>2</sub> [582], BTa <sub>2</sub> [583], BTi <sub>2</sub> [584], BW <sub>2</sub> [585], Ba[586], BeTa <sub>2</sub> [587], Bi <sub>3</sub> In <sub>4</sub> Pb[588] Ca <sub>5</sub> Pt <sub>3</sub> [589], Ca <sub>5</sub> Si <sub>3</sub> [590] Cs <sub>9</sub> InO <sub>4</sub> [591], GaHf <sub>2</sub> [592], GaPt <sub>3</sub> [593], GaZr <sub>2</sub> [594], Ga <sub>5</sub> Pd[595], GeHf <sub>2</sub> [596], Ge <sub>3</sub> Mo <sub>5</sub> [597] Ge <sub>3</sub> Nb <sub>5</sub> [598], Ge <sub>3</sub> W <sub>5</sub> [599], Nb <sub>5</sub> SiSn <sub>2</sub> [600] Nb <sub>5</sub> Si <sub>3</sub> [601], Pb <sub>3</sub> Sr <sub>5</sub> [602], PdTl <sub>2</sub> [603], SZr <sub>2</sub> [604] Si <sub>3</sub> W <sub>5</sub> [605], Sr[606], Te <sub>3</sub> Tl <sub>5</sub> [607]
163	AlF <sub>6</sub> LiPd[608], F <sub>6</sub> GaLiPd[609], In <sub>2</sub> Pt <sub>3</sub> [610]
164	AlCl <sub>3</sub> [611], CSc <sub>2</sub> [612], CaHg <sub>2</sub> [613], CaSb <sub>2</sub> Zn <sub>2</sub> [614], Cl <sub>2</sub> Ti[615], HfTe <sub>2</sub> [616, 617], Hg <sub>7</sub> K <sub>2</sub> [618] Hg <sub>7</sub> Rb <sub>2</sub> [618], OTi <sub>2</sub> [619], Sb <sub>2</sub> SrZn <sub>2</sub> [620], Se <sub>2</sub> Ti[621]
166	BaPb <sub>3</sub> [622], Be <sub>17</sub> Hf <sub>2</sub> [623], Be <sub>17</sub> Nb <sub>2</sub> [624], Be <sub>17</sub> Ta <sub>2</sub> [623], GeTe[625, 626], Pb <sub>2</sub> Pd <sub>3</sub> S <sub>2</sub> [627] Pb <sub>2</sub> Pd <sub>3</sub> Se <sub>2</sub> [628], Po[629], SbSn[630] AgInSe <sub>2</sub> [631], As <sub>3</sub> NaZn <sub>4</sub> [632], Bi <sub>4</sub> Te <sub>3</sub> [633], C <sub>3</sub> Al <sub>8</sub> N <sub>4</sub> [634], Cd <sub>4</sub> KP <sub>3</sub> [635], GeP <sub>3</sub> [636], In <sub>2</sub> Se <sub>3</sub> [637], Pd <sub>3</sub> Se <sub>2</sub> Tl <sub>2</sub> [638], Ag <sub>4</sub> Sb <sub>2</sub> Sr[639]
167	Al <sub>2</sub> Mo[640], Cs <sub>8</sub> Ga <sub>11</sub> [641], F <sub>3</sub> Mo[642], F <sub>3</sub> Pd[643], F <sub>3</sub> Ti[644], In <sub>11</sub> K <sub>8</sub> [645] K <sub>8</sub> Tl <sub>11</sub> [646], Mo <sub>9</sub> S <sub>11</sub> Tl <sub>2</sub> [647], NPd <sub>3</sub> [648]

## List 489 good topological Semimetal

176	AgCd <sub>3</sub> F <sub>20</sub> Hf <sub>3</sub> [649], AgCd <sub>3</sub> F <sub>20</sub> Zr <sub>3</sub> [649], AgMo <sub>3</sub> Se <sub>3</sub> [650], AsNb <sub>3</sub> Te <sub>3</sub> [651] CsMo <sub>3</sub> S <sub>3</sub> [652], CsMo <sub>3</sub> Se <sub>3</sub> [653], CsMo <sub>3</sub> Te <sub>3</sub> [654], InMo <sub>3</sub> Se <sub>3</sub> [655, 656] InMo <sub>3</sub> Te <sub>3</sub> [655, 656], KMo <sub>3</sub> S <sub>3</sub> [655], KMo <sub>3</sub> Se <sub>3</sub> [655, 656], KMo <sub>3</sub> Te <sub>3</sub> [656, 657], LiMo <sub>3</sub> Se <sub>3</sub> [658] Mo <sub>3</sub> NaSe <sub>3</sub> [655, 656], Mo <sub>3</sub> NaTe <sub>3</sub> [656, 658], Mo <sub>3</sub> RbS <sub>3</sub> [655], Mo <sub>3</sub> RbSe <sub>3</sub> [658, 659] Mo <sub>3</sub> RbTe <sub>3</sub> [660], Mo <sub>3</sub> Se <sub>3</sub> Tl[656, 658], Mo <sub>6</sub> Se <sub>6</sub> Tl <sub>2</sub> [661], NbSe <sub>2</sub> [662]
187	AgN[663], CLiNaO <sub>3</sub> [664], CMo[665], CPt[666], CTa[666], CW[667] InS <sub>2</sub> Ta[668], InSe <sub>2</sub> Ta[669], MoP[670], NNb[671], NPd[672], NW[673] NZr[674], N <sub>2</sub> Pt[675], N <sub>2</sub> Ta[675], NbS[676], STa[677] STi[678], TeZr[679]
188	I <sub>3</sub> LiSc[680]
189	AlHfPt[681], AlPdY[682], AlPtZr[681], Al <sub>2</sub> Hf <sub>6</sub> Pt[683], AsPd <sub>2</sub> [684] AuInY[685], GaPtZr[681], GePdSc[686], GePd <sub>2</sub> [687], GePt <sub>2</sub> [688], InPdY[689] InPtSc[690], InPtY[691], MgPdY[692], NTa[693], PdTiY[694], PdYZn[695] Pd <sub>2</sub> Si[696], PtSb <sub>2</sub> Zr <sub>6</sub> [697], Pt <sub>2</sub> Si[698]
190	AlHfPt[699], AlPtZr[700], B <sub>4</sub> Ga <sub>3</sub> Pt <sub>9</sub> [701], GaHfPd[702], GaPtZr[703], Li <sub>2</sub> Sb[704], PdScSn[705]
191	Au <sub>2</sub> Ba[706], Au <sub>5</sub> K[707], Au <sub>5</sub> Rb[707], B <sub>2</sub> Mo[708], BaGa <sub>2</sub> [709], BaPd <sub>5</sub> [710], Be <sub>2</sub> Hf[711] Be <sub>5</sub> Hf[712], Be <sub>5</sub> Zr[713], CaGa <sub>2</sub> [714], CaHg <sub>2</sub> [715], Hf[716], Hf <sub>2</sub> N <sub>3</sub> Ta[717], Hg[718] LiNNa <sub>2</sub> [719], NNa <sub>3</sub> [720], NOTa[721], NTa[722], N <sub>3</sub> TaTi <sub>2</sub> [723], N <sub>3</sub> TaZr <sub>2</sub> [723], N <sub>4</sub> Ta <sub>2</sub> [724], S <sub>2</sub> Ti[725]
193	AgIn <sub>3</sub> Zr <sub>5</sub> [726], AgPb <sub>3</sub> Zr <sub>5</sub> [727], AgSb <sub>3</sub> Zr <sub>5</sub> [728], AlSn <sub>3</sub> Zr <sub>5</sub> [729], Al <sub>3</sub> Hf <sub>5</sub> [730] Al <sub>3</sub> Hf <sub>5</sub> N[731], Al <sub>3</sub> Ta <sub>5</sub> [732], Al <sub>3</sub> Zr <sub>5</sub> [733], Al <sub>4</sub> Zr <sub>5</sub> [734] AsPb <sub>3</sub> Zr <sub>5</sub> [735], AsSb <sub>3</sub> Zr <sub>5</sub> [736], AsSn <sub>3</sub> Zr <sub>5</sub> [737], As <sub>3</sub> Ca <sub>5</sub> [738], As <sub>3</sub> Sr <sub>5</sub> [739] BNb <sub>5</sub> Si <sub>3</sub> [740], BSn <sub>3</sub> Zr <sub>5</sub> [741], Ba <sub>3</sub> N[742], Ba <sub>5</sub> Bi <sub>3</sub> [743] Ba <sub>5</sub> Sb <sub>3</sub> [743], Bi <sub>3</sub> Sr <sub>5</sub> [744], CSb <sub>3</sub> Zr <sub>5</sub> [745], CSn <sub>3</sub> Zr <sub>5</sub> [746], Ca <sub>5</sub> Sb <sub>3</sub> [747], CdPb <sub>3</sub> Zr <sub>5</sub> [748] Cl <sub>3</sub> Ti[749, 750], Cl <sub>3</sub> Zr[750, 751], GaSn <sub>3</sub> Zr <sub>5</sub> [752], Ga <sub>3</sub> Hf <sub>3</sub> Nb <sub>2</sub> [753] Ga <sub>3</sub> Nb <sub>5</sub> [754], Ga <sub>3</sub> Sc <sub>5</sub> [755], Ga <sub>3</sub> Ta <sub>5</sub> [756], Ga <sub>4</sub> Nb <sub>5</sub> [757], Ga <sub>4</sub> Ti <sub>5</sub> [758] Ga <sub>4</sub> Zr <sub>5</sub> [758], GePb <sub>3</sub> Zr <sub>5</sub> [759], Ge <sub>3</sub> Mo <sub>5</sub> [760], Ge <sub>3</sub> Nb <sub>5</sub> [761] Ge <sub>3</sub> Sc <sub>5</sub> [762], Ge <sub>3</sub> Ta <sub>5</sub> [763], Ge <sub>3</sub> Y <sub>5</sub> [764], Hf <sub>3</sub> [765, 766], Hf <sub>5</sub> Sb <sub>3</sub> Zn[767], Hf <sub>5</sub> Si <sub>3</sub> [768] Hf <sub>5</sub> Sn <sub>3</sub> [769], Hf <sub>5</sub> Sn <sub>4</sub> [770], Hg <sub>3</sub> Mg <sub>5</sub> [771], I <sub>3</sub> Nb[772] I <sub>3</sub> Ti[773, 774], I <sub>3</sub> Zr[775], InPb <sub>3</sub> Zr <sub>5</sub> [776], K <sub>3</sub> Nb <sub>5</sub> O <sub>21</sub> [777] Mo <sub>5</sub> Si <sub>3</sub> [778], NSn <sub>3</sub> Zr <sub>5</sub> [779], N <sub>6</sub> Nb <sub>5</sub> [780], N <sub>6</sub> Ta <sub>5</sub> [781] Nb <sub>5</sub> OPt <sub>3</sub> [782], Nb <sub>5</sub> PSi <sub>3</sub> [783], Nb <sub>5</sub> Si <sub>3</sub> [784], PPb <sub>3</sub> Zr <sub>5</sub> [785] PSn <sub>3</sub> Zr <sub>5</sub> [786], P <sub>3</sub> Ti <sub>5</sub> [787], Pb <sub>3</sub> SbZr <sub>5</sub> [788], Pb <sub>3</sub> Sc <sub>5</sub> [789] Pb <sub>3</sub> SeZr <sub>5</sub> [788], Pb <sub>3</sub> SiZr <sub>5</sub> [788], Pb <sub>3</sub> SnZr <sub>5</sub> [788], Pb <sub>3</sub> Y <sub>5</sub> [790] SSb <sub>3</sub> Zr <sub>5</sub> [791], SSn <sub>3</sub> Zr <sub>5</sub> [792], Sb <sub>3</sub> SiZr <sub>5</sub> [793] Sb <sub>3</sub> Sr <sub>5</sub> [794], Sb <sub>4</sub> Zr <sub>5</sub> [795], SeSn <sub>3</sub> Zr <sub>5</sub> [796], SiSn <sub>3</sub> Zr <sub>5</sub> [796] Si <sub>3</sub> Ta <sub>5</sub> [797], Si <sub>3</sub> Y <sub>5</sub> [798], Sn <sub>4</sub> Zr <sub>5</sub> [799]

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# Symmetry indicator

## ➤ TaAs family

*Weng, Fang, Fang, Bernevig, & Dai, PRX (2015)*

*Huang et al., Nature Commun. (2015)*

## ➤ Magnetic

# Conclusion

□ We propose a highly efficient method to explore topological materials

**What one need to do is to look the expansion coefficient!**

□ We propose numbers of new topological materials.



**Thank you very much for your  
attention**

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# 当前的问题和可能的方案

## **Na<sub>3</sub>Bi**

Wang, Sun, Chen, Franchini, Xu, Weng, Dai, and Fang, PRB (2012).

## **Cd<sub>3</sub>As<sub>2</sub>**

Wang, Weng, Wu, Dai, and Fang, PRB (2013)

## **SnTe**

Hsieh, Lin, Liu, Duan, Bansil and Fu, Nat. Commun. (2012).

Wang, West, Liu, Li, Yan, Gu, Zhang, and Duan, PRB (2014).